



# Some aspects of reactive thermo-chemically non-equilibrium plasmas

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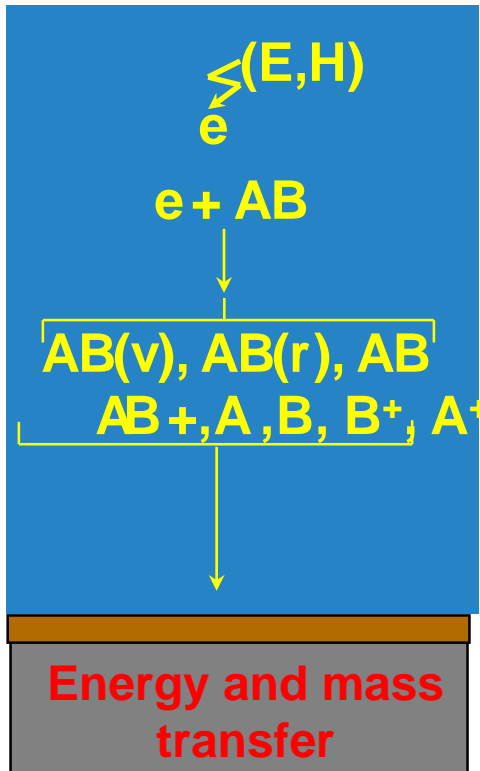
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# Overview

- Phenomenological description
- Electron kinetics in non-equilibrium discharges
- Reactivity and kinetics of electronically excited states– consequence on ionization kinetics and plasma characteristics : the example of hydrogen
- Vibrational kinetics and its impact on molecular dissociation : the example of molecular nitrogen
- Molecular growth, i.e., “plasma polymerization” in non equilibrium discharges :
  - Sputtering DC discharges with graphite electrodes
  - Ar/C<sub>2</sub>H<sub>2</sub> capacitively coupled RF discharges
- Interplay between power coupling, energy relaxation, chemistry, and transport in reactive plasma
  - H<sub>2</sub>/CH<sub>4</sub> microwave cavity plasmas
  - Ar/C<sub>2</sub>H<sub>2</sub> capacitively coupled RF discharges
- Few slides on the methodology

# Few words on laboratory plasmas : a phenomenological description



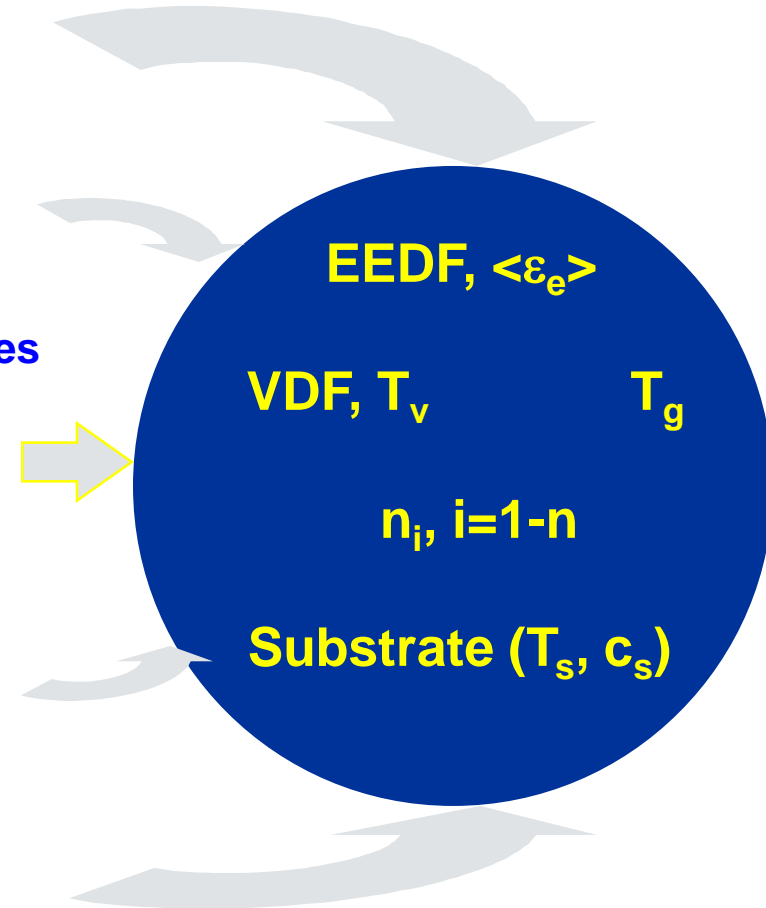
Plasma-wave interaction  
Electron heating

Electron-heavy species collisions  
Energy transfer, ionisation, etc

Collisions between heavy species  
Energy transfer, chemistry, clustering,

charged particle transport :  
Drift, Diffusion, convection  
Energy and mass transport

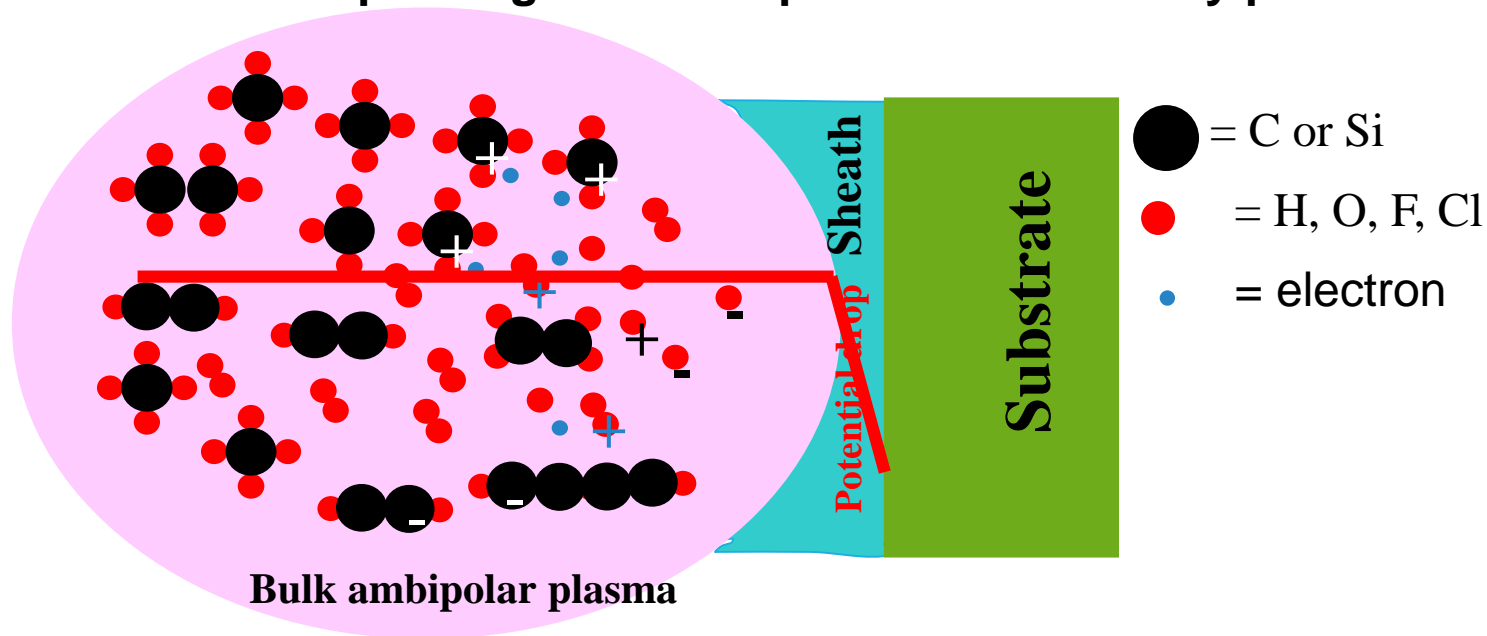
plasma/surface Interaction  
Energy and mass transfer



# Few words on laboratory plasmas:

## different types of chemical species and practical interest

### Gas phase generated species in laboratory plasmas



#### Gas phase reactions :

- **Atomic/molecular** species : O-atom, F-atom, H-atom
- Radicals :  $\text{CH}_3$ ,  $\text{CF}_3$ ,  $\text{SiH}_2$ ,  $\text{SiH}$ ,  $\text{SiH}_3$
- Positive ions :  $\text{CH}_5^+$ ,  $\text{SiH}_5^+$ ,  $\text{H}_2^+$ ,  $\text{H}_3^+$ ,  $\text{H}^+$ ,  $\text{O}_4^+$ , etc.
- Negative ions :  $\text{H}^-$ ,  $\text{O}^-$
- Large molecular structures :  $\text{S}_n\text{H}_m^-$ ,  $\text{C}_n\text{H}_m^-$ , etc.
- Nucleation and growth of solid particles

#### Plasma processes

- Surface treatment
- Material deposition
- Chemical synthesis/decomposition
- Particle/nanostructure nucleation
- Active species/photons /fast particles sources

# Few words on Laboratory plasmas

## Phenomenological description

Plasma-wave interaction  
**Electron heating**

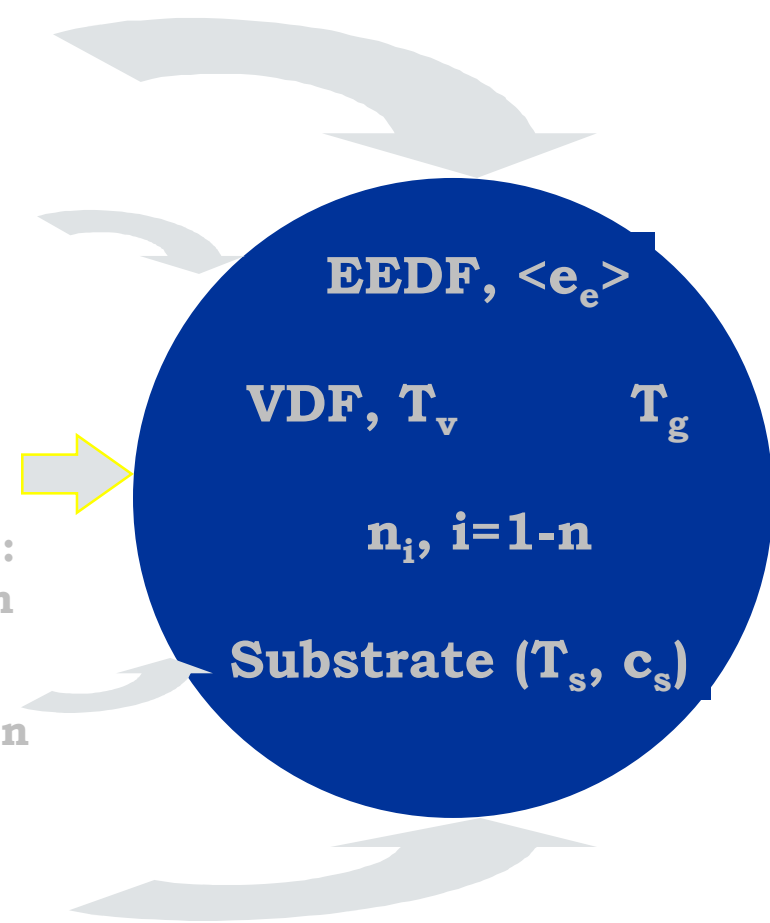
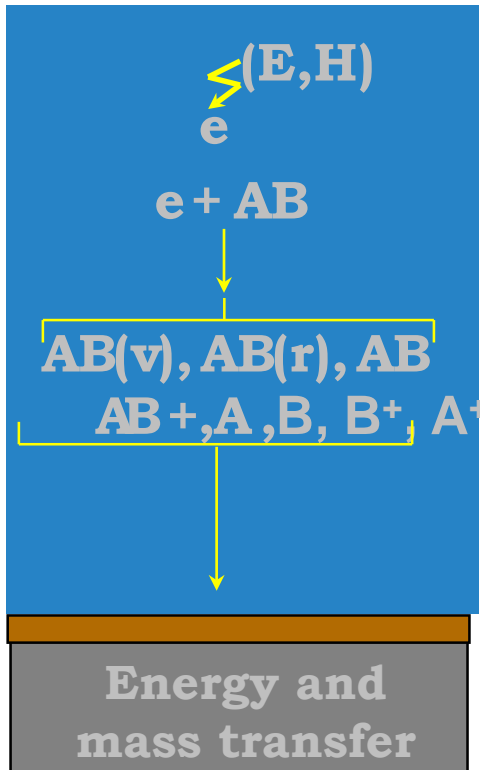
**Electron-heavy species collisions**  
**Energy transfer, ionisation, etc**

Collisions between heavy species  
 Energy transfer,

chemistry, clustering,  
 charged particle transport :  
 Drift, Diffusion, convection

Energy and mass transport  
 plasma/surface Interaction

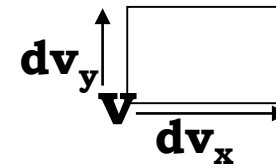
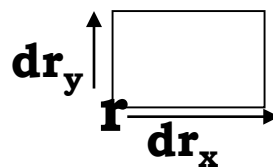
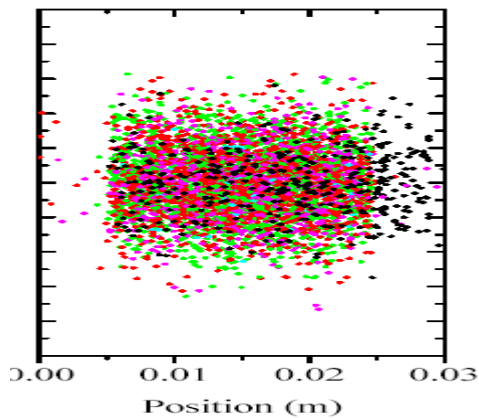
Energy and mass transfer



**Electrons are the drivers for the plasma reactivity → Electron kinetics**

# Electron probability density function

Electrons occupy different positions and show different velocity values **Probability Density Function in phase space :  $f_e(v_e, r_e)$**   
 $f_e(v_e, r_e) dv_e dr_e$  : number of electrons at positions in the position range  $r_e \pm \frac{dr_e}{2}$  with velocity-values in the range  $v_e \pm \frac{dv_e}{2}$



**Local electron density**

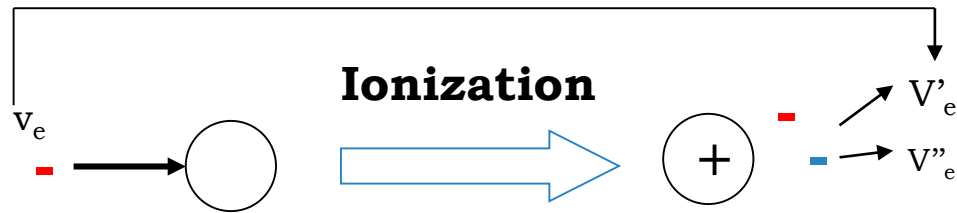
$$n_e(r_e) = \int_{v_e} \frac{d^2 N_e}{dr_e} = \int_{v_e} f_e(v_e, r_e) dv_e$$

Electron states in a plasma : **electric field** , **collisions**, **transport/boundaries**

$$\frac{\partial f_e}{\partial t} + \underbrace{\vec{v}_e \cdot \vec{\nabla}}_{\text{green}} f_e - \frac{e}{m} \underbrace{\vec{E} \cdot \vec{\nabla}}_{\text{red}} f_e = \underbrace{C(f_e)}_{\text{blue}}$$

# Collision frequency and rate constant

## e-HS Collisions : Local and instantaneous



Collision frequency for 1 e<sup>-</sup> in s<sup>-1</sup> :  $\nu_c(v_e) = \sigma_c(v_e)v_e n_{HS}$

A collision is characterized by an integral collision cross section  $\bar{\sigma}_c(v_e)$  in cm<sup>2</sup>

For the whole electron population the collision frequency per unit volume s<sup>-1</sup>cm<sup>-3</sup>:

$$\bar{\nu}_c = \int_{v_e} f(v_e) dv_e v_e \sigma_e n_{tot} = k_c n_e n_{tot}$$

Also define the rate constant

$$k_c = \frac{\int_{v_e} f(v_e) dv_e v_e \sigma_e}{\int_{v_e} f(v_e) dv_e}$$

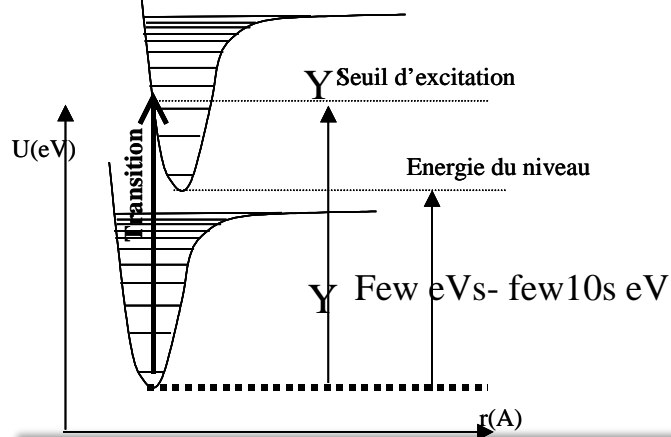
$$\int_{v_e} f(v_e) dv_e = n_e$$

# Different types of Electron/heavy species Collisions

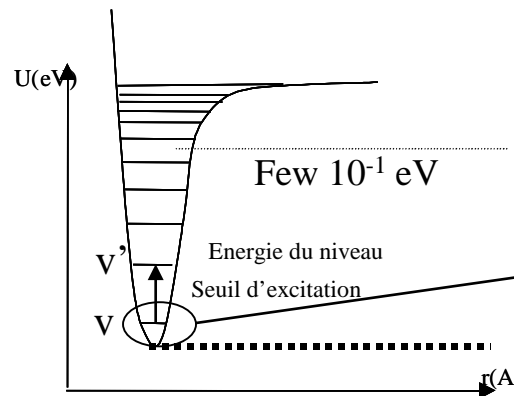
**Elastic:** => e-HS Momentum and kinetic energy transfer (electron current and direct gas heating):  $e^-(\mathbf{v}_e) + A_2(\mathbf{v}_{HS}) \Rightarrow e^-(\mathbf{v}'_e) + A_2(\mathbf{v}'_{HS})$  **Total kinetic energy (e+HS) conserved**

**Inelastic non-reactive** => **Excitation/de-excitation of the electronic, vibrational or rotational modes and electron cooling(superelastic)**

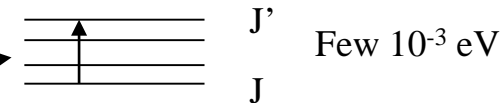
## Electronic excitation



## Vibrational excitation



## Rotational excitation



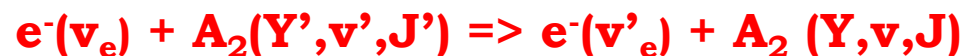


# Different types of Electron/heavy species Collisions

**Reactive inelastic** → Ionization, dissociation, dissociative attachment

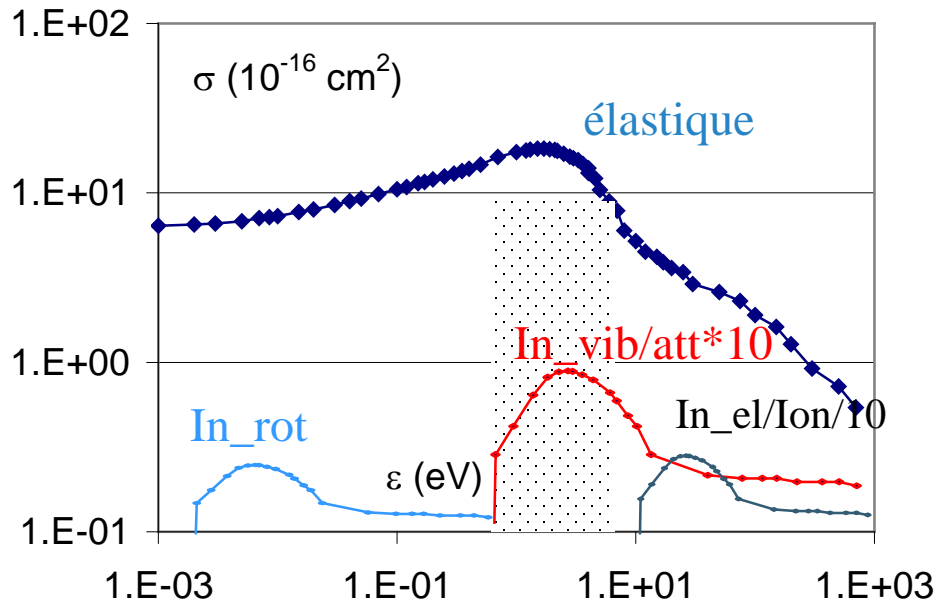


**Super-elastic** (inelastic reverse process) → internal mode de-excitation and electron heating)



# Cross section trends, collision frequency and energy transfer rate

## Typical trends



## Collision frequency

$$\nu_c(\nu_e) = \sigma_c(\nu_e)\nu_e n_{HS}$$

**Elastic**  $\gg$   **$In_{vib} \geq In_{e\_other}$**

## Energy transfer rate:

**Elastic** :  $\nu_{E-el}(\nu_e) \propto \sigma_c(\nu_e)\nu_e n_{HS} \frac{m_e}{m_{HS}} \epsilon_e$

**Inelastic**:  $\nu_{E-In}(\nu_e) \propto \sigma_c(\nu_e)\nu_e n_{HS} \epsilon_{th}$

**$In_{vib} \geq In_{e\_other} \gg El$**

# E-HS rate constant and electron PDF

Major questions when investigating reactive non equilibrium plasmas for a population of electrons :

1/ What are the rates of the primary e-HS collisions that sustain the plasma and generate the primary active species ?

→ **Chemistry**

$$r_{e-HS} = k_{e-HS} n_e n_{HS}$$

2/ What are the rates and the main channels for electron energy dissipation in the gas (self-consistent model) ?

→ **Electron energy dissipation**

$$r_{e-HS}^e = k_{e-HS} n_e n_{HS} \varepsilon_{th}$$

$$k_{e-HS}(r_e) = \int_{v_e} f(r_e, v_e) \sigma_{e-HS}(v_e) v_e dv_e$$

$$\int_{v_e} f(r_e, v_e) dv_e = 1$$

**Boltzmann equation**

$$\frac{\partial f_e}{\partial t} + \vec{v}_e \cdot \vec{\nabla} f_e - \frac{e}{m_e} \vec{E} \cdot \vec{\nabla}_{v_e} f_e = C(f_e)$$

# Electron PDF and solution of the BE

- Three major practical situations :

1- **No assumption** : Solve the Boltzmann in the full phase space. Suitable for low density plasma, i.e.,  $n_e < 10^{12} - 10^{13} \text{ cm}^{-3}$  – no electron-electron collisions. Necessary when non local and anomalous effects (anisotropy, fluctuation, beam-like, etc.) are important : sheaths, magnetized plasmas, etc

2- **Assume that the PDF is local (2a) and isotropic(2b)** : develop the PDF as a LC of spherical harmonic functions and keep either only the isotropic component (classical/old) or the isotropic + the first anisotropy (newer see Hageaar and Pitchford PSST) : suitable for most of the reactive plasmas.

3- Assume a local Maxwellian PDF suitable for large density-low pressure plasmas (**enhanced electron-electron collisions** → **Maxwell distribution**)

$$f_e(v_e) = \left( \frac{m_e}{2\pi kT_e} \right)^{\frac{3}{2}} \exp\left( -\frac{m_e v_e^2}{2kT_e} \right)$$

# Illustration case 1 : statistical particle simulation

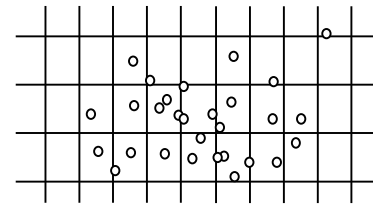
## Electron dynamics and Particle In Cell simulation

First : charged particles (electrons) dynamics  
(Particle In Cell, no collisions)

Work with  $N$  *super-particles*

$$N_{s,r\acute{e}el} = \alpha_s \cdot N_{s,num}$$

Distributed on a space grid



$$\Delta x \leq 0,3\lambda_D$$



Space-resolve the sheaths

For each superparticles

**Time-integrate momentum equations on  $\Delta t$**

$$v(t + \Delta t) = v(t) + \int_{\Delta t} \frac{ez}{m} E dt$$

$$x(t + \Delta t) = x(t) + \int_{\Delta t} v dt$$

$$\Delta t \leq \frac{\Delta x}{v_{\max}}$$

Interpolation on  
superparticle positions

For each plasma cell  
Compute the charge  
density

$$\rho_j = \frac{1}{V_c} \sum_v \lambda_v Q_v$$

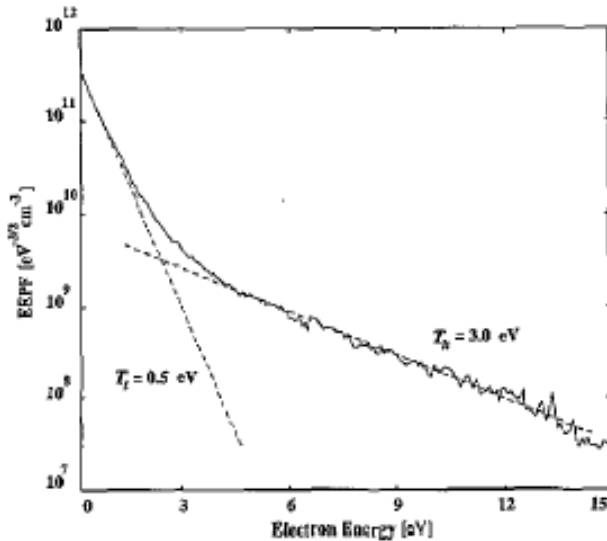
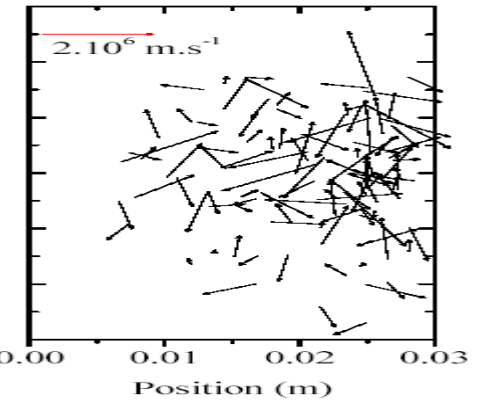
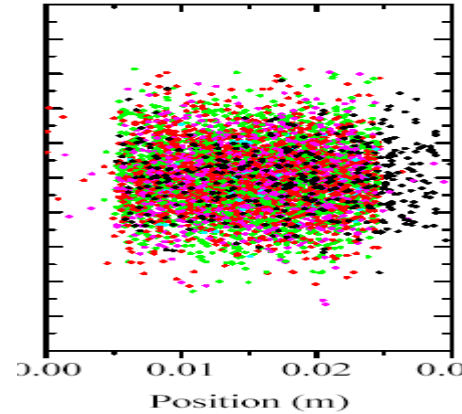
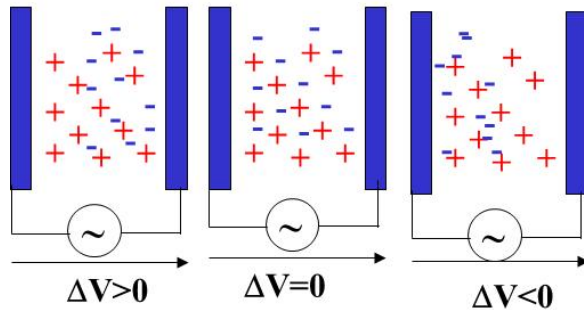
**Compute the selfconsistent EM field**

Maxwell or Poisson's



# Illustration case 1: Capacitively coupled RF discharge in N<sub>2</sub>

N<sub>2</sub>  
0.1 torr  
V<sub>RF</sub> = 100 V  
Gap = 3 cm



## Stochastic heating at the sheath edge

→ Bi-Maxwellian EEDF

→ Large temperature 3 eV for high energy electrons

→ Low temperature, 0.5 eV, for low energy electrons

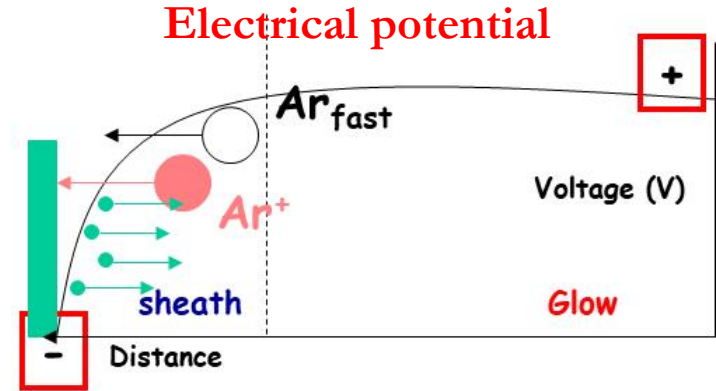
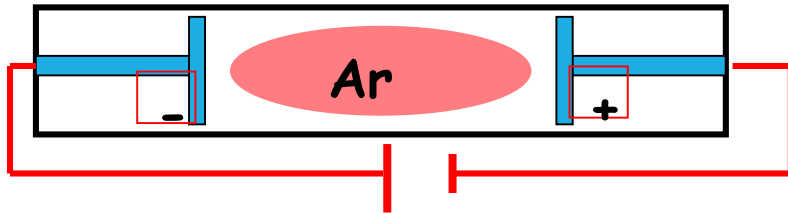
$$k_{n-max}/k_{max} > 10^2$$

S. Longo, K. Hassouni, D. Iasillo and M. Capitelli, *J. Phys. III, Fr*, **7** (1997) 707-718 DOI: 10.1051/jp3:1997133

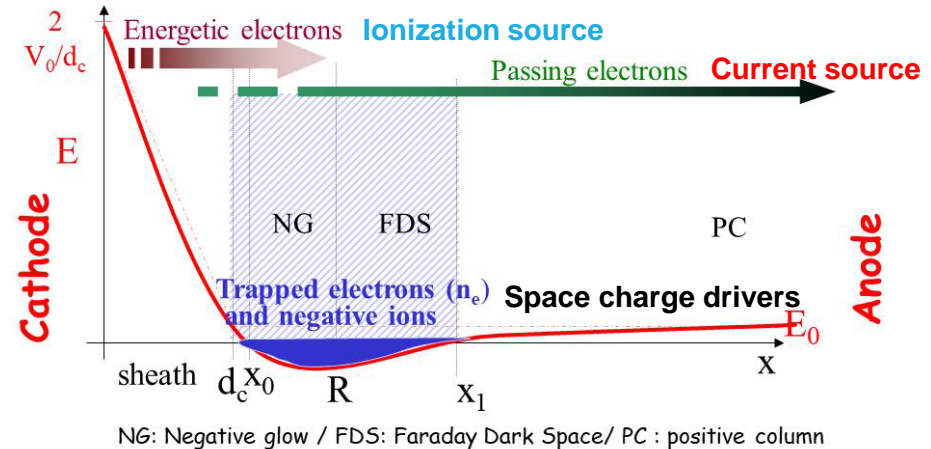
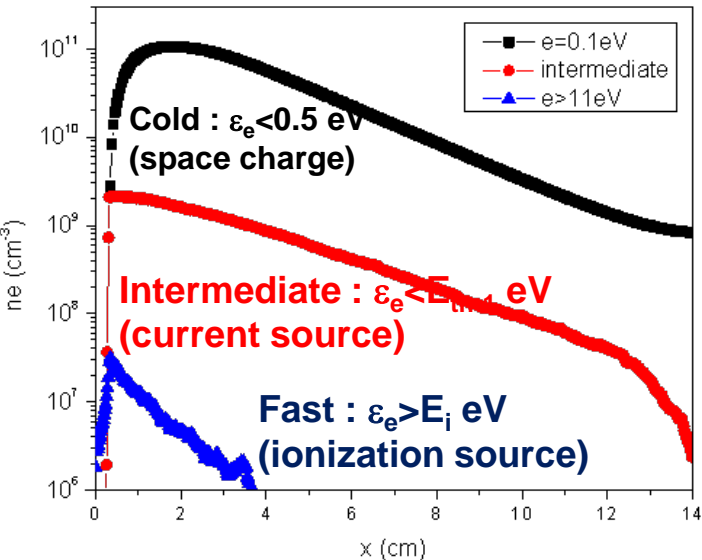
S. Longo, M. Capitelli and K. Hassouni, *Phys. IV France* **07** (1997) C4-271-C4-281 DOI: 10.1051/jp4:1997422

# Illustration case 1 : electron kinetics in argon DC discharge (Monte Carlo)

## DC discharge



## Three electron populations



A. Michau et al. Plasma Chem. And Plasma Proc., **32**, 451-470(2012). Doi : [10.1007/s11090-012-9357-0](https://doi.org/10.1007/s11090-012-9357-0)

A. Michau et al. PSST, **25**, (2016), paper# 015019(16 pp) doi:10.1088/0963-0252/25/1/015019PI



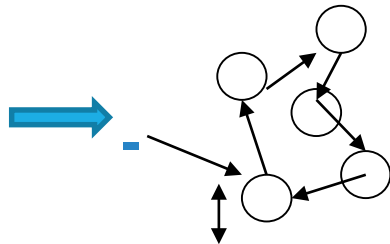
# Illustration case 2 : electron kinetics in moderate pressure

## H<sub>2</sub>/CH<sub>4</sub> plasmas

Plasmas dominated by collisions and fulfilling:

- **Locality** :  $\vec{v}_e \cdot \vec{\nabla} f_e$  negligible
- **Isotropy** :  $v_x \Leftrightarrow v_y \Leftrightarrow v_z \rightarrow f(v_e) = F(e_e)$

Collisions induce isotropic velocity distribution



Simplification

The pdf,  $f_e$ , is approximated by the isotropic component,  $F_e$ , in its spherical harmonic expansion

$$\frac{\partial f_e}{\partial t} - \vec{v}_e \cdot \vec{\nabla} f_e - \frac{e}{m_e} \vec{E} \cdot \vec{\nabla}_v f_e = C(f_e)$$

Much simpler BE

$$\frac{\partial F_e}{\partial t} + \frac{\partial}{\partial \varepsilon_e} J_E + \frac{\partial}{\partial \varepsilon_e} J_{el} = In(F_e) + Sup(F_e)$$

What about electric-field modulation ( $E = E_0 \cos(\omega_{HF} t)$ ) ?

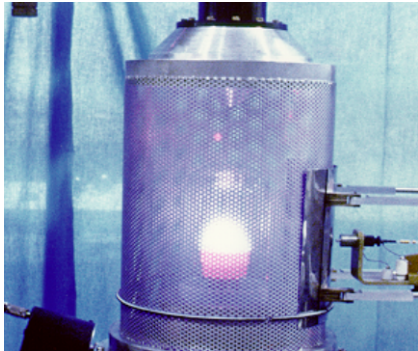
=> electron may follow the field depending on  $v_{e-ex}/\omega_{HF}$  => 2 options

1- Low pressure case :  $\frac{m_e}{m} v_{el} + v_{in} \ll \omega_{HF} \rightarrow$  stationary situation

$\rightarrow$  Use the effective field assumption

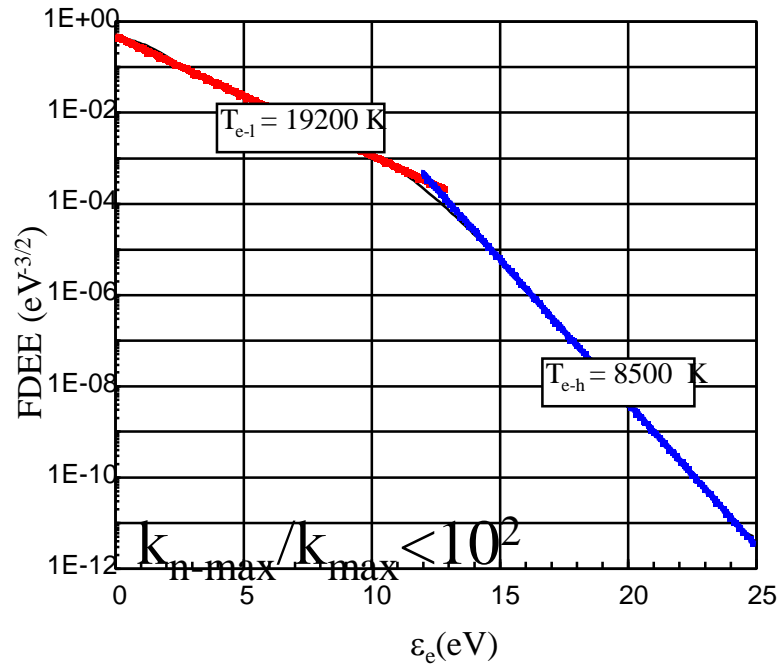
$$E_{\text{eff}} = \frac{E_0}{\sqrt{2}} \frac{v_m^e}{(v_m^e{}^2 + \omega^2)^{1/2}}$$

2- high pressure case :  $\frac{m_e}{m} v_{el} + v_{in} \gg \omega_{HF} \rightarrow$  non-stationary situation for electrons  $\rightarrow$  time-integrate the BE



# EEDF behavior in collisional high frequency plasmas

## Bimodal stationary distribution

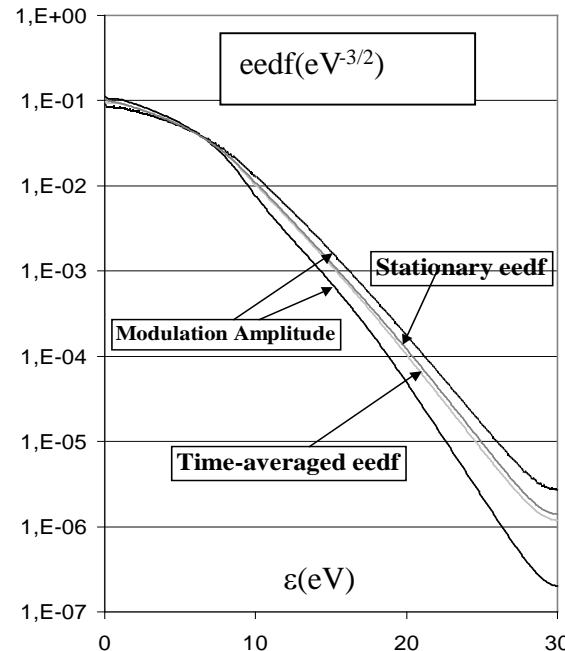


**EEDF Bimodal  $T_{e-h} = f(T_{e-l})$  (univoque)**  
**Energy balance  $\rightarrow T_{e-l}$**

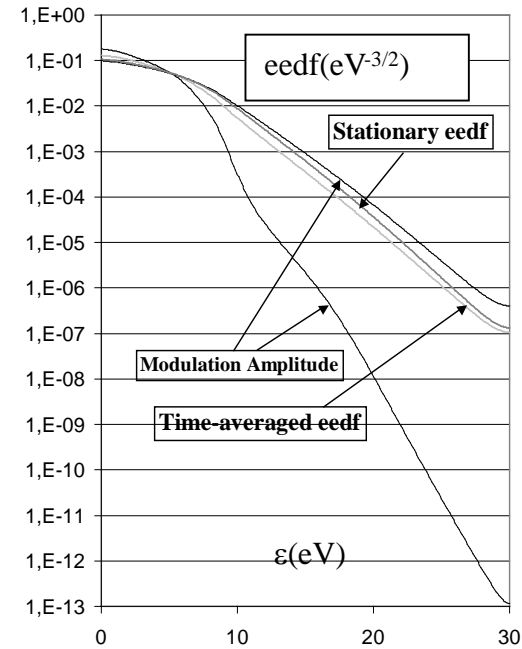
- K. Hassouni et al. PSST **8**(1999), 494-512
- K. Hassouni et al. Surf. Coat. Technol. **97**(1-3), 391-403 (1997)
- M. Capitelli et al. Plasma Chem. And Plasma Proc. **16**, pages 153–171 (1996)

## Impact of the effective field assumption

### Low pressure case



### High pressure case

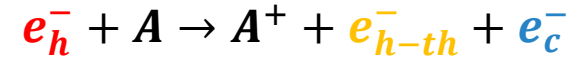
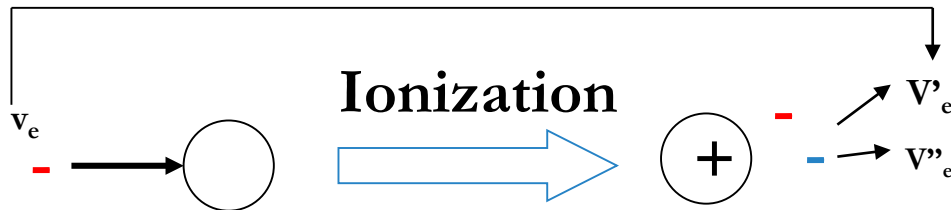


**Effective field assumption and non stationary model give almost the same result**

- M. Capitelli et al. Phys. Rev. E **1843** (1996)
- W. Morscheidt et al. Plasma Chem. Plasma Proc. **23**, 117-140, (2003) doi: [10.1023/A:1022472904111](https://doi.org/10.1023/A:1022472904111)

# Discharge maintenance : Ionization kinetics

In principle : the most straightforward mechanism → **direct ionization** :

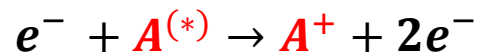


But less straightforward mechanism may exist : multi-step

- **Penning**



- **step-wise**



# Stepwise and Penning mechanism illustration : The coupling between electronic and ionization kinetics in H<sub>2</sub> plasmas

Chemical species : H<sub>2</sub> and H

Direct Ionization :  $e^- + H_2 \rightarrow H_2^+ + 2e^-$  and  $e^- + H \rightarrow H^+ + 2e^-$

## Closer look

### - Electronic excitation of : H<sub>2</sub> and H

- $e^- + H_2 \Rightarrow e^- + H_2^*$
- $e^- + H(n) \Rightarrow e^- + H(m)$

### - Stepwise ionization involving electronically excited states

- $e^- + H_2 = [H_2^{**}] \Rightarrow H_2^+ + 2e^-$
- $e^- + H(n) \Rightarrow 2e^- + H^+$

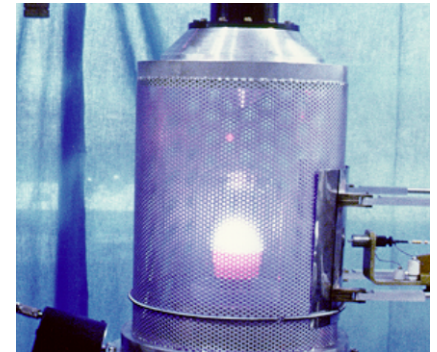
### - Penning ionization through H-atom excited state :

- $H(n) + H_2 \Rightarrow [H_3^{++} + e^-]$  ou  $[3H]$

### - Need to look at collisional and radiative relaxation processes

- $H_2^* (+ M) \Rightarrow H_2^{**}$  or  $2H (+M)$
- $H(n) + H \Rightarrow H(m) + H$
- $M^* \Rightarrow M^{*'} + h\nu$  (M=H or H<sub>2</sub>)

H<sub>2</sub> Microwave plasmas  
Microwave cavity coupling  
Pressure : few tens of mbars



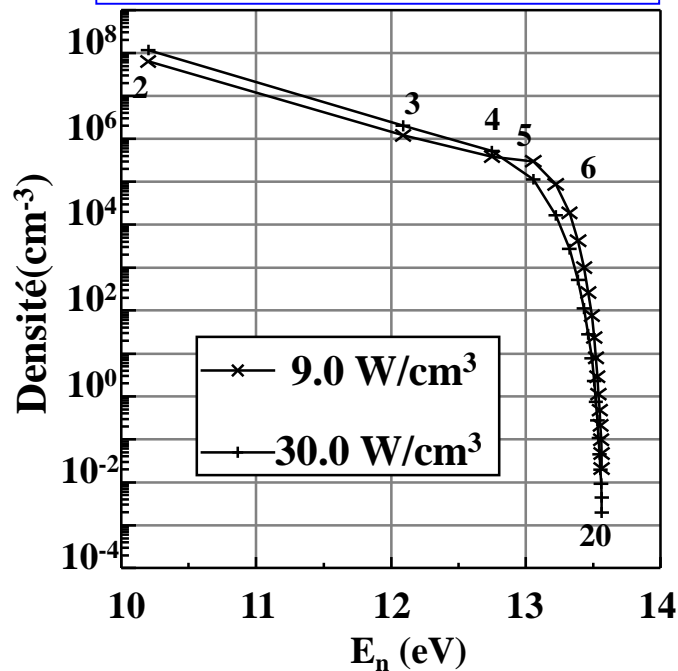
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K. Hassouni et al. Surf. Coat. Technol. **97**(1-3), 391-403 (1997)

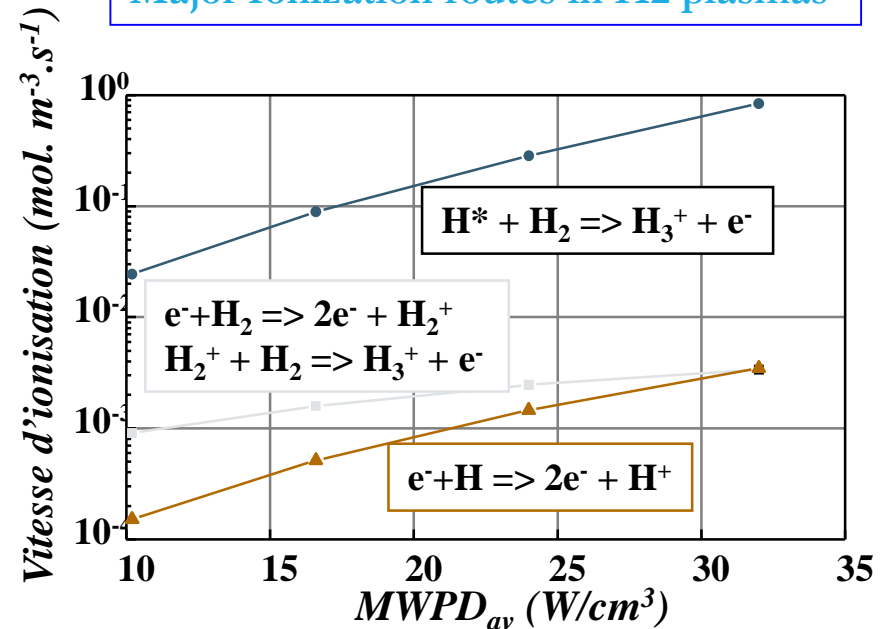
# Stepwise and Penning mechanisms illustration : The coupling between electronic and ionization kinetics in H<sub>2</sub> plasmas

Prime importance → ionization kinetics sustains the discharge

Electronically excited state distribution for H-atom



Major Ionization routes in H<sub>2</sub> plasmas



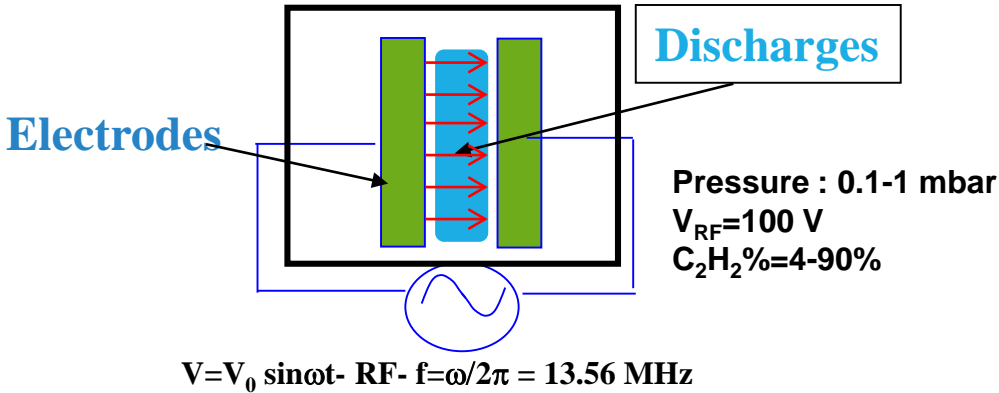
K. Hassouni et al. PSST **8**(1999), 494-512

**Ionization is mainly governed by H(n=2-3) quenching**

K. Hassouni et al. PSST **8**(1999), 494-512

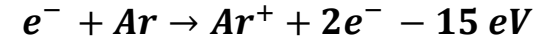
K. Hassouni et al. Surf. Coat. Technol. **97**(1-3), 391-403 (1997)

# Penning vs stepwise vs direct ionization mechanisms : the example of CCRF Ar/C<sub>2</sub>H<sub>2</sub>



## Ionization routes

### 1- Direct on Argon :

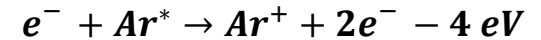


### 2- Stepwise on argon and Penning on C<sub>2</sub>H<sub>2</sub>

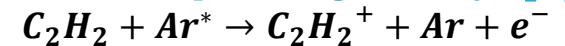
#### 2-1. excitation of argon metastable



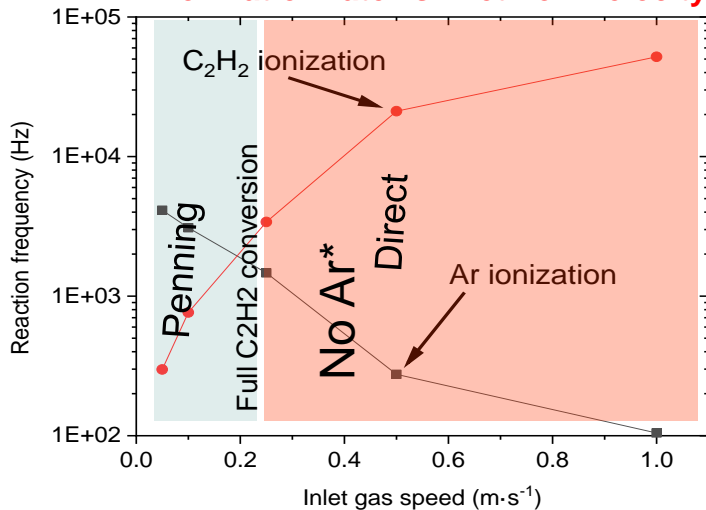
#### 2-2-a- Ionization of Ar\*



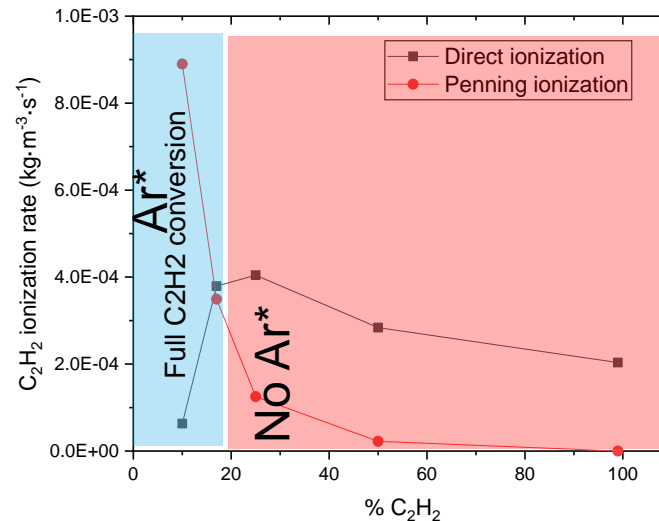
#### 2-2-b- Collisional quenching of Ar\* by C<sub>2</sub>H<sub>2</sub>



### Ionization rate vs inlet flow velocity



### Ionization rate vs acetylene content

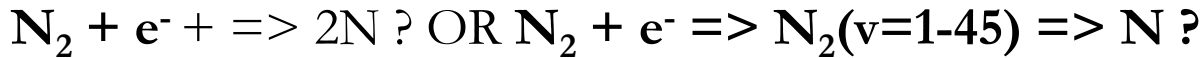


**Ar\* populated**  
**→ Penning >> direct**

G. Tetard et al. PSST (2021), 30, paper# 105015(22 pp) doi 10.1088/1361-6595/ac2a17

# Primary activation processes : the coupling between molecular dissociation and vibrational kinetics in N<sub>2</sub> plasmas

What's the main dissociation mechanism in the plasma bulk :

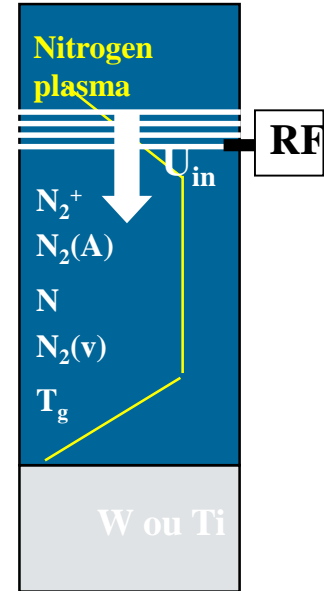


How much N at the discharge exit ?

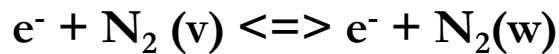
Plug flow model

$$\frac{dn_s}{dt} - W_s = 0 \quad z = U_{in} t$$

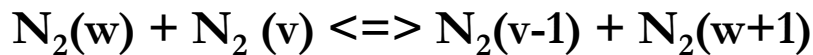
$T_g = \text{cste}$  ; Maxwellian eedf et  $T_e = \text{cste}$



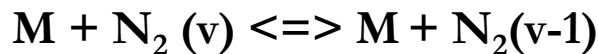
e-v excitation processes



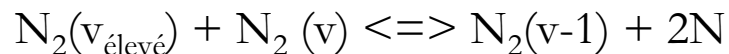
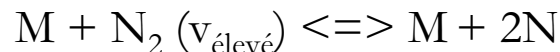
Resonant v-v relaxation processes



V-t de-excitation process



dissociation

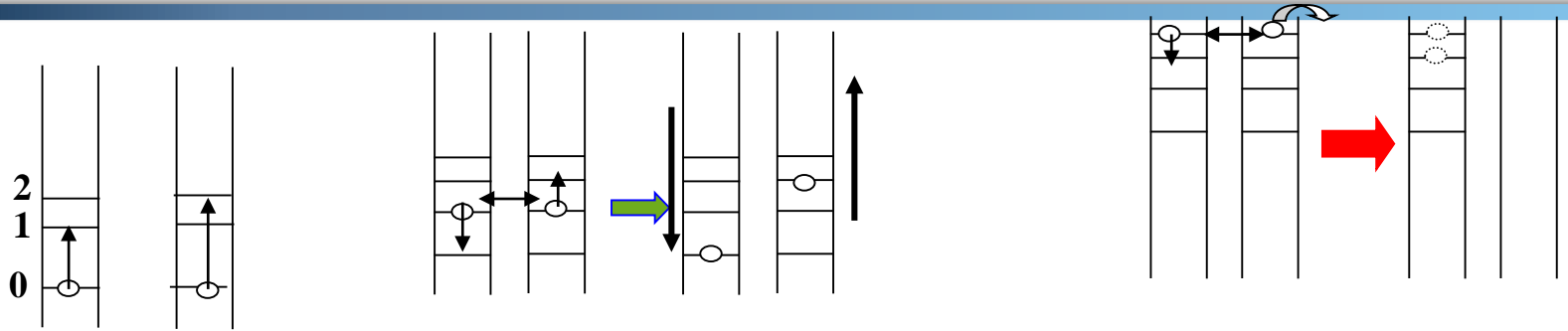


K. Hassouni et al. Computer and Chem. Eng. (1993), 17(Supp. 1), S505-S510

S. Longo et al. , J. Phys. III, Fr, 7 (1997) 707-718 DOI: 10.1051/jp3:1997133

S. Longo et al. . Phys. IV France 07 (1997) C4-271-C4-281 DOI: 10.1051/jp4:1997422

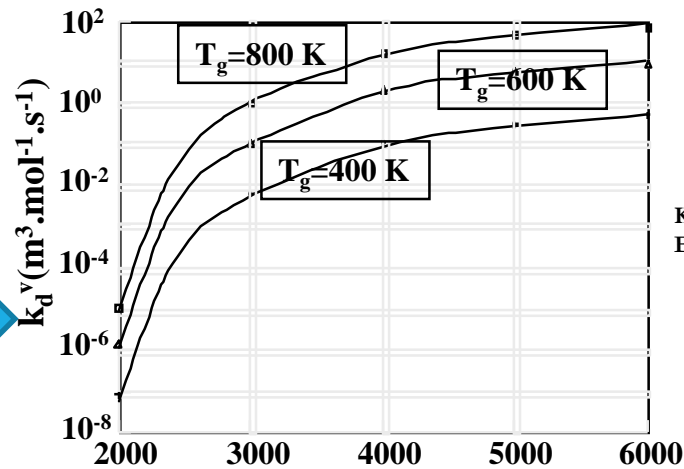
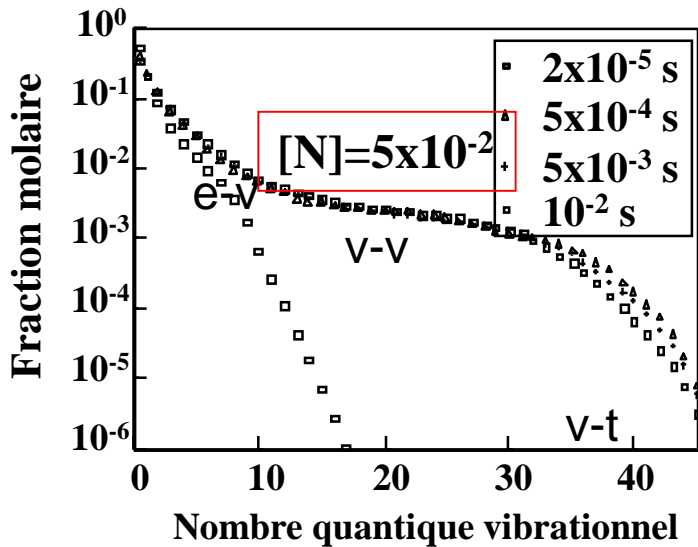
# Primary activation processes : the coupling between molecular dissociation and vibrational kinetics in N<sub>2</sub> plasmas



$e^- + N_2(v=0) \Rightarrow e^- + N_2(v=1-4)$   
pompage sur les bas niveaux

$N_2(v) + N_2(w) \Rightarrow N_2(v-1) + N_2(w+1)$   
Montée dans l'échelle vibrationnelle

$N_2(v_l) + N_2(v) \text{ (ou M)} \Rightarrow 2N + N_2(v-1) \text{ (ou M)}$   
Dissociation du dernier niveau par collision avec un lourd

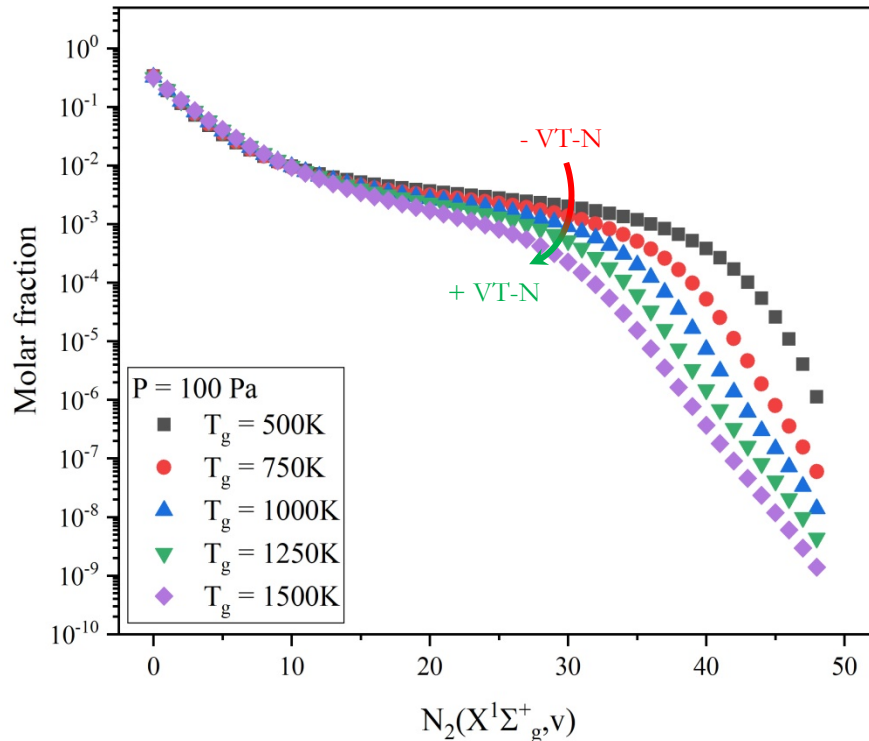


K. Hassouni et al. Computer and Chem. Eng. (1993), 17(Supp. 1), S505-S510

$(n_e T_e) \rightarrow T_v$  THEN  $(T_v T_g) \rightarrow$  dissociation kinetics :  $2N_2 \Rightarrow 2N + N_2$  [ $k_d^v = f(T_v, T_g)$ ]

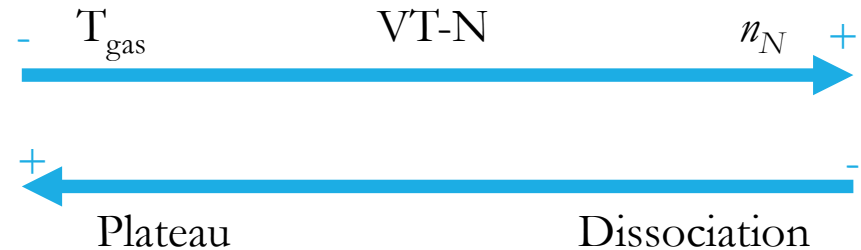


# Self-limiting effects for the vibrational dissociation mechanism in N<sub>2</sub> plasmas



Maxwellian  
Low plateau  
VT-N processes

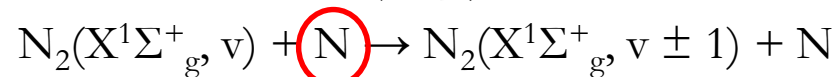
Non-Maxwellian  
Large plateau  
VV relaxation



Self-limiting mechanism of N-atoms to molecular dissociation via v-D mechanism

How to enhance N-atom production?

VT-N:

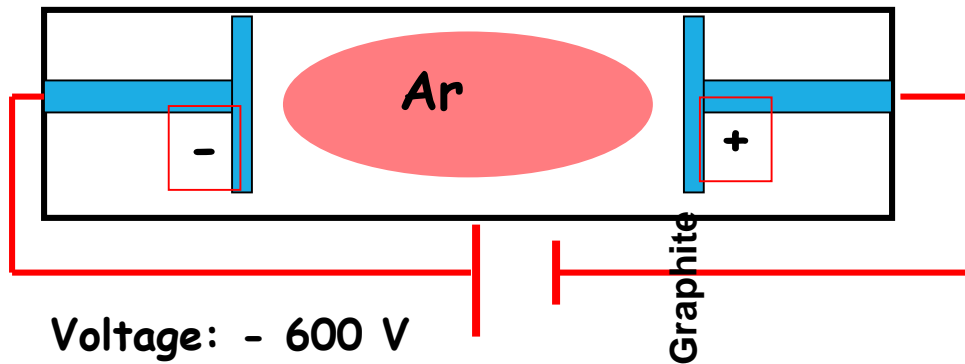


**By pulsing the discharge!**

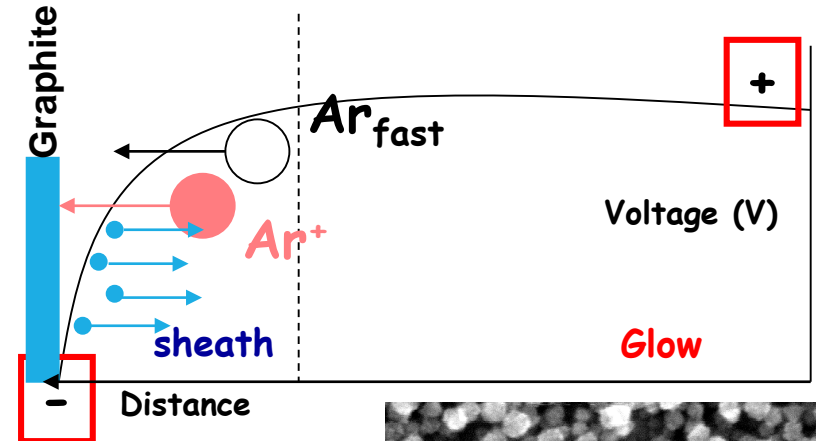
S. Prasanna et al. to be submitted to PoP

# Attachment and molecular growth kinetics in non equilibrium plasmas : the case of sputtering DC discharges

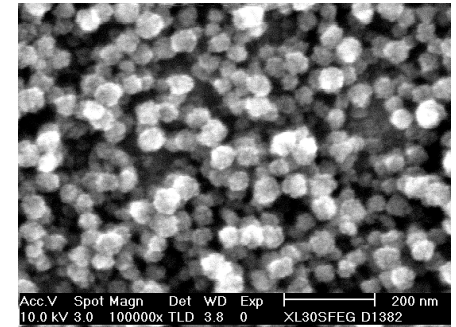
## DC discharge



Voltage: - 600 V  
 Gap distance: 14 cm  
 Electrode diameter: 5 cm  
 Current:  $8 \times 10^{-2}$  A



C. Dominique and C. Arnas,  
 J. Appl. Phys. (2007) 101(12),  
 10.1063/1.2748365



Dusts are produced by the sputtering of the graphite cathode:

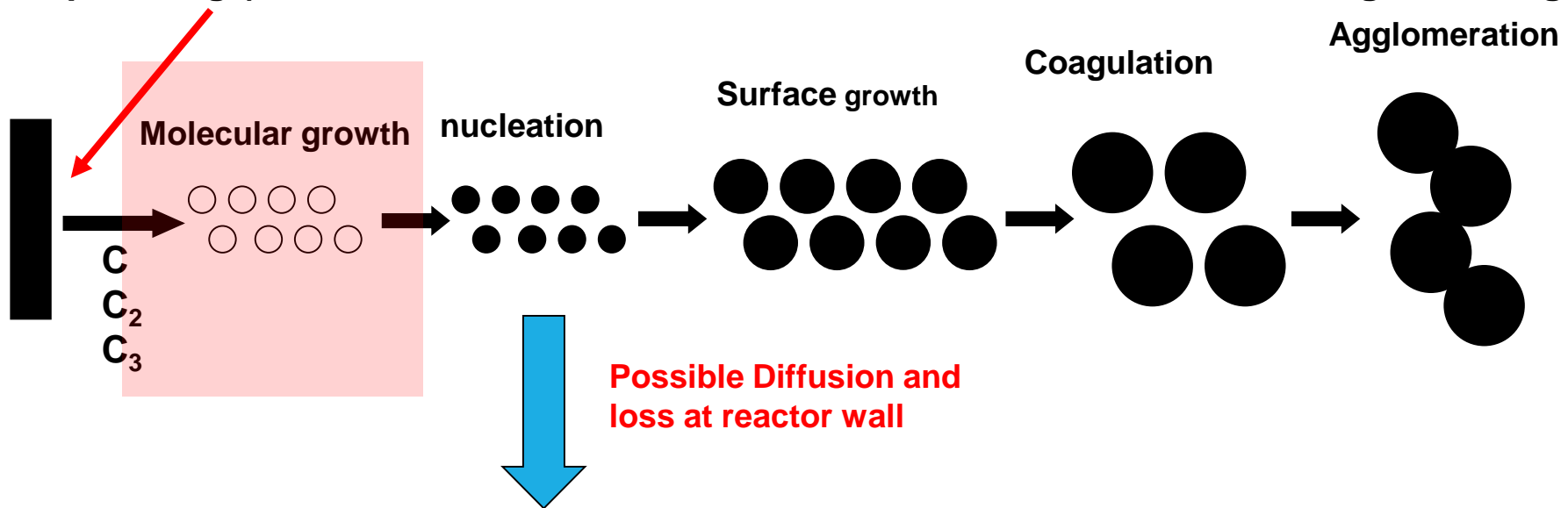
- Argon ions accelerated in the sheath

- Fast neutrals resulting from charge transfer  $Ar_{slow} + Ar^+ \rightarrow Ar^+ + Ar_{fast}$

What is the mechanism responsible for dust formation ?

# Attachment and molecular growth kinetics in non equilibrium plasmas : the case of sputtering DC discharges

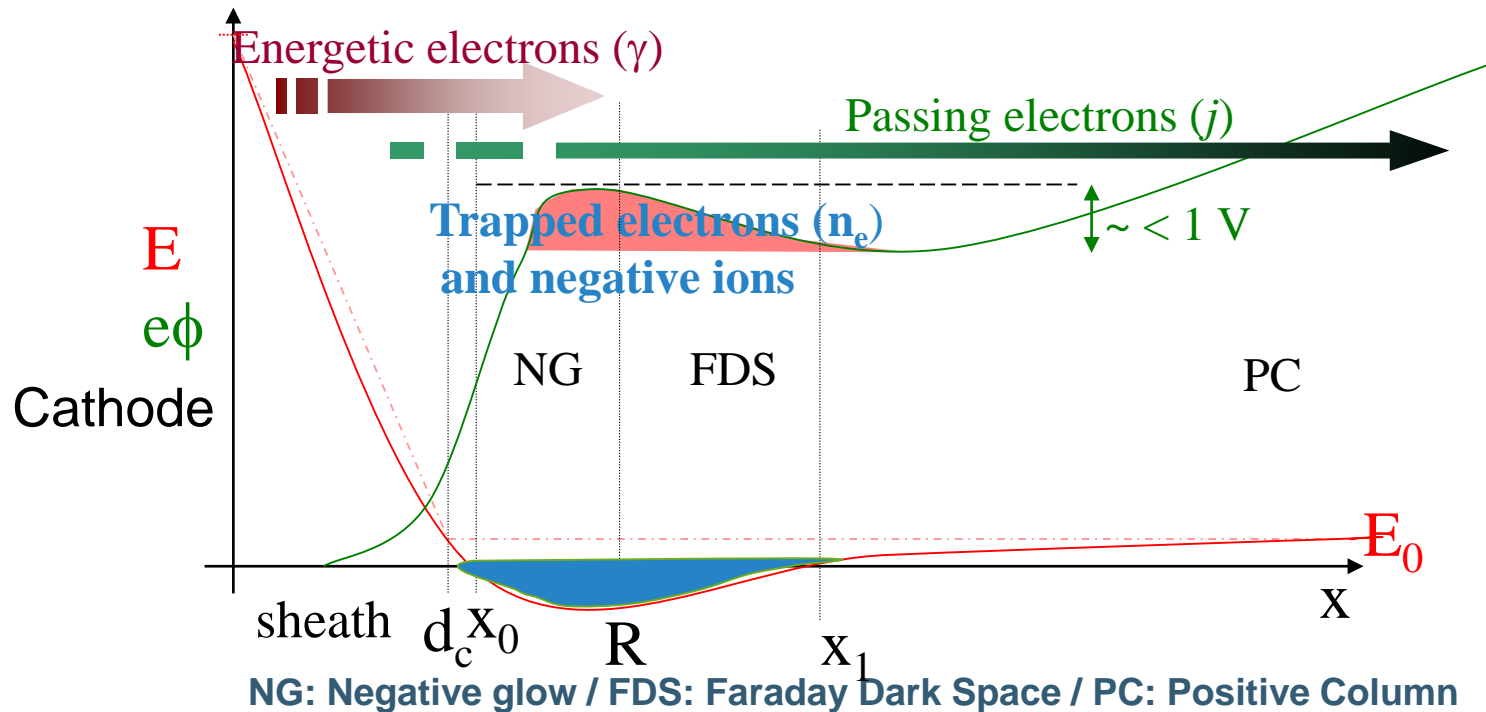
$\text{Ar}^+ + \text{Ar}$  sputtering ( Ion accelerated in the sheath – Fast neutral created from charge exchange)



$T_{\text{growth}} \sim 100 \cdot T_{\text{diffusion-neutral}}$   
→ Neutral cannot grow  
→ charged species : electrostatic trapping ?

# Attachment and molecular growth kinetics in non equilibrium plasmas : Confining effect of field reversal in DC discharges

Three electron populations: **fast** (ionizing), **intermediate** (energy below the first excitation threshold), **cold** (trapped in the field reversal)

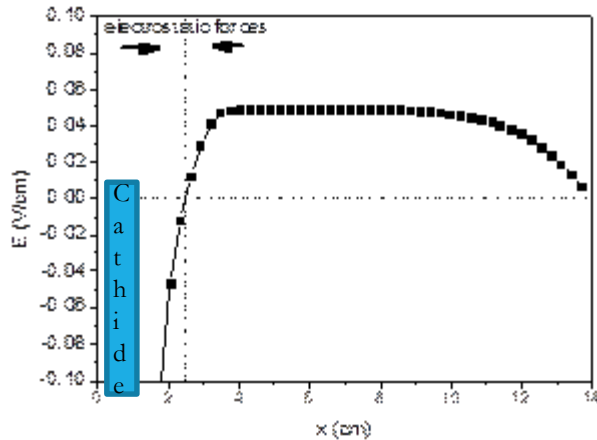


## Existence of a field reversal

→ electrical potential structure is confining for negatively charged species

# Attachment and molecular growth kinetics in non equilibrium plasmas : field reversal and molecular growth in DC discharges

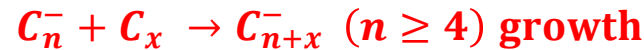
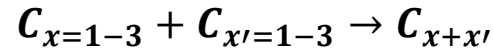
## Electric field reversal



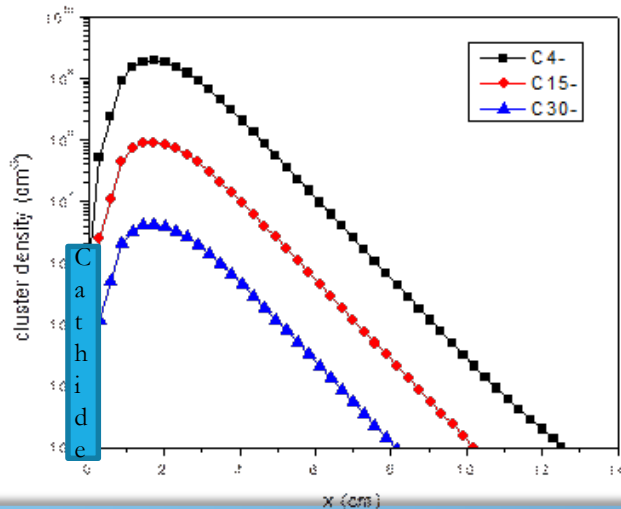
**Field-reversal at 2 cm from the cathode.**

**→ Electrostatic trapping of negative species is possible**

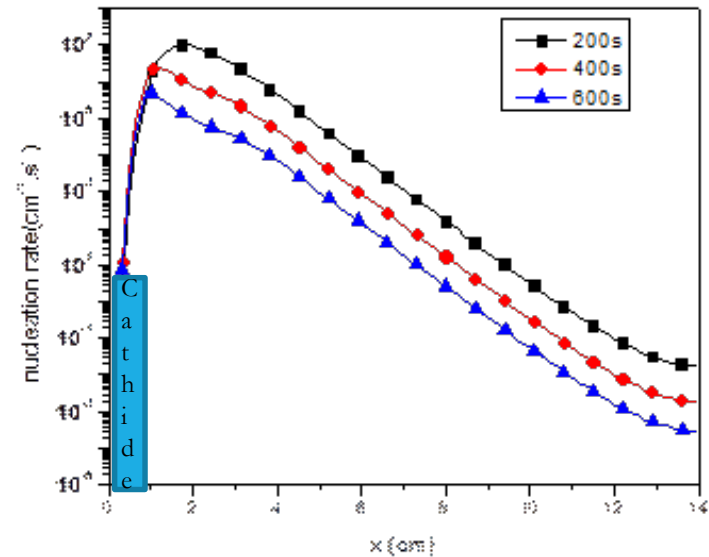
**sputtered C, C<sub>2</sub>, C<sub>3</sub> induce a molecular growth :**



## Negatively charged clusters



## Nucleation rate



A. Michau et al. Plasma Chem. And Plasma Proc., **32**, 451-470(2012). Doi : 10.1007/s11090-012-9357-0

A. Michau et al. PSST, **25**, (2016), paper# 015019(16 pp) doi:10.1088/0963-0252/25/1/015019PI

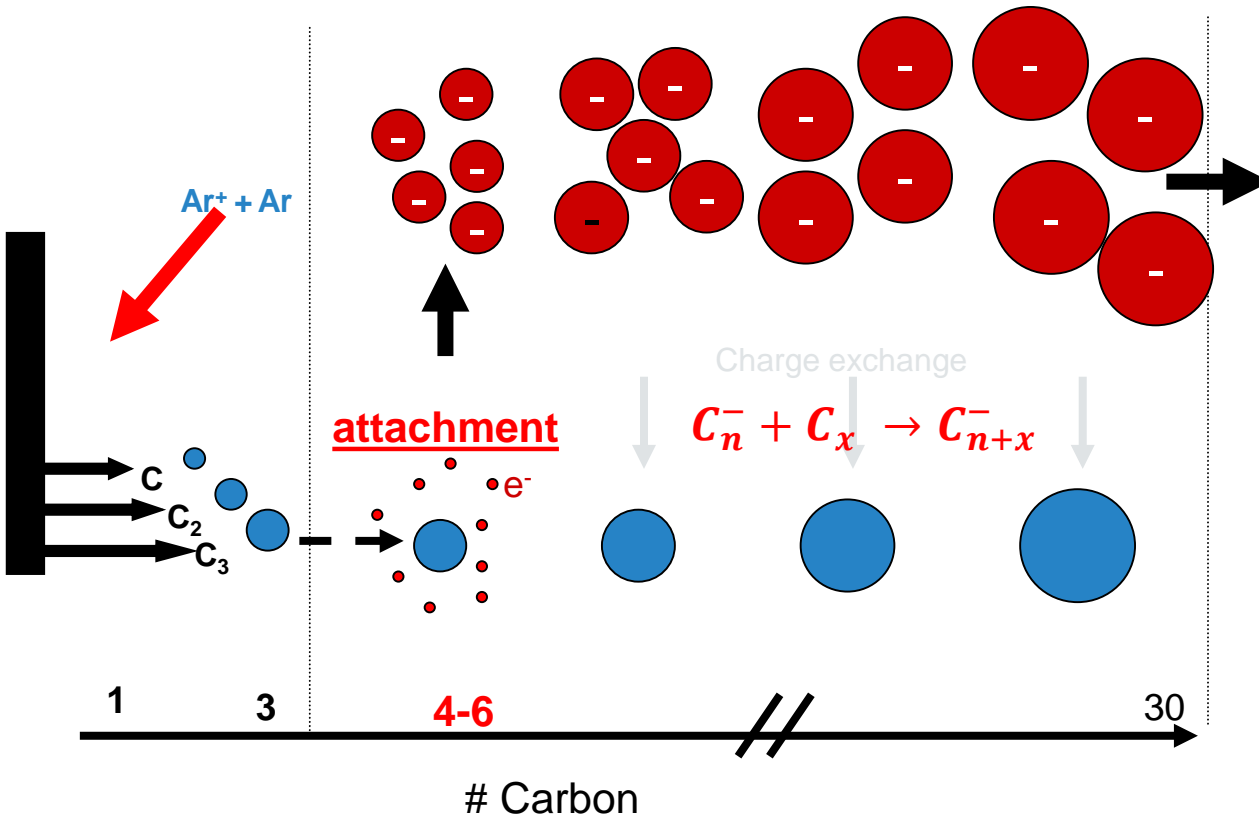
A. Michau et al. PSST (2010) paper „ 034023(7pp) doi 10.1088/0963-0252/19/3/034023

# Attachment and molecular growth kinetics in sputtering DC discharge

Sputtering

Molecular growth

Nucleation



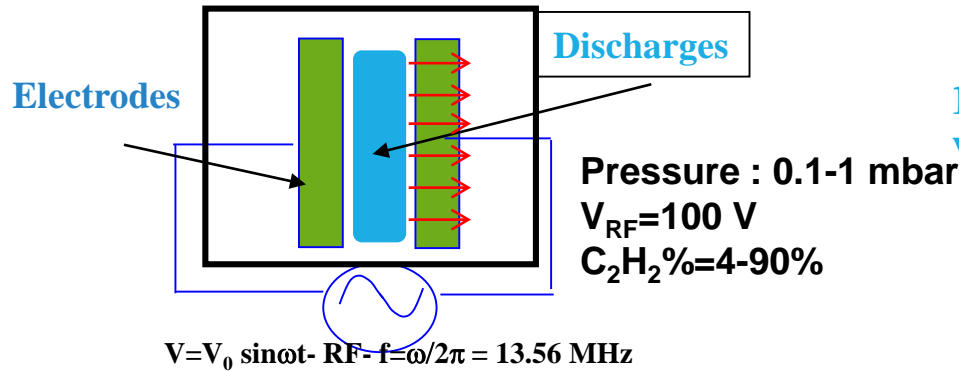
A. Michau et al. Plasma Chem. And Plasma Proc., **32**, 451-470(2012). Doi : 10.1007/s11090-012-9357-0

A. Michau et al. PSST, **25**, (2016), paper# 015019(16 pp) doi:10.1088/0963-0252/25/1/015019PI

A. Michau et al. PSST (2010) paper „ 034023(7pp) doi 10.1088/0963-0252/19/3/034023

Key-role of Field reversal and negatively charged species for molecular growth

# Ionization and molecular growth kinetics in Ar/C<sub>2</sub>H<sub>2</sub> CCRF plasmas



Dust formation is often (if not always) observed  
What is the mechanism for particle nucleation ?

Answer :

- RF plasmas fluid model (G. Tetard et al. PSST (2021), **30**, paper# 105015(22 pp) doi 10.1088/1361-6595/ac2a17)

Handle large time-scale chemistry

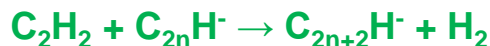
- Growth model (from M. Mao et al., J. Phys D (2008) 225201(14 pp))



$$k \sim 3 \cdot 10^7 \text{ m}^3 \cdot \text{mol}^{-1} \cdot \text{s}^{-1}$$



$$k \sim 4 \cdot 10^7 \text{ m}^3 \cdot \text{mol}^{-1} \cdot \text{s}^{-1}$$

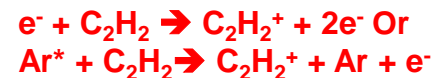


$$k \sim 6 \cdot 10^5 \text{ m}^3 \cdot \text{mol}^{-1} \cdot \text{s}^{-1}$$



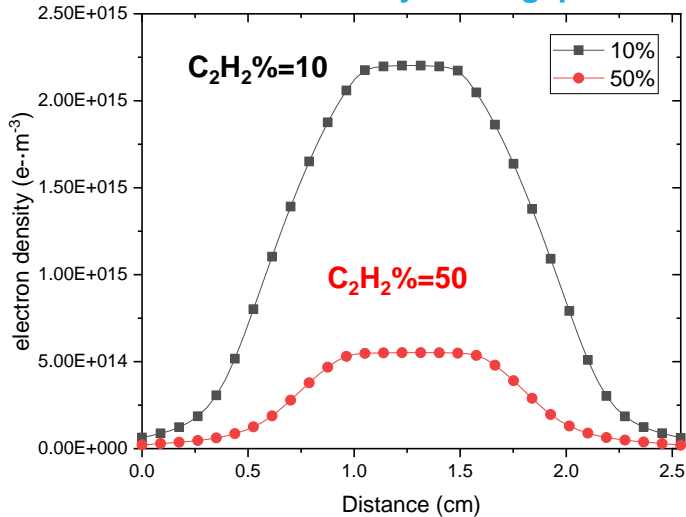
$$k \sim 6 \cdot 10^7 \text{ m}^3 \cdot \text{mol}^{-1} \cdot \text{s}^{-1}$$

The growth starts with

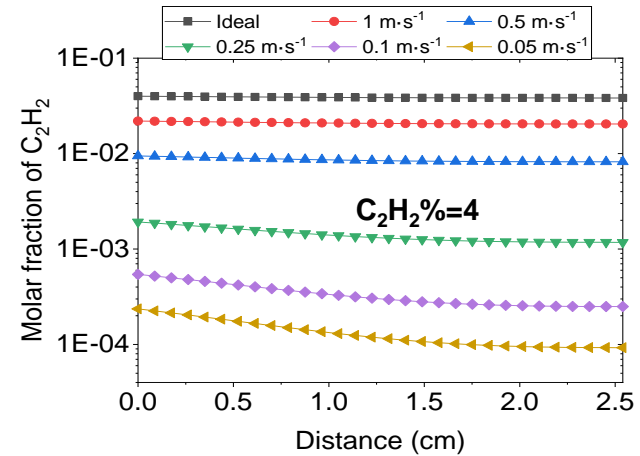


# Ionization and molecular growth kinetics in Ar/C<sub>2</sub>H<sub>2</sub> CCRF plasmas : discharge structure and composition

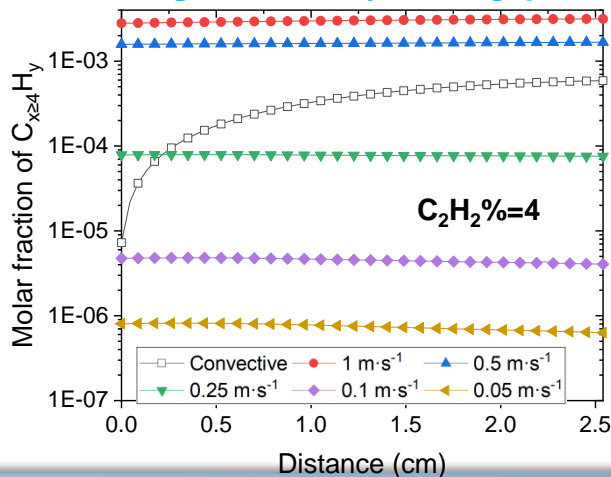
Electron density in the gap



C<sub>2</sub>H<sub>2</sub> density in the gap



Large HC density in the gap

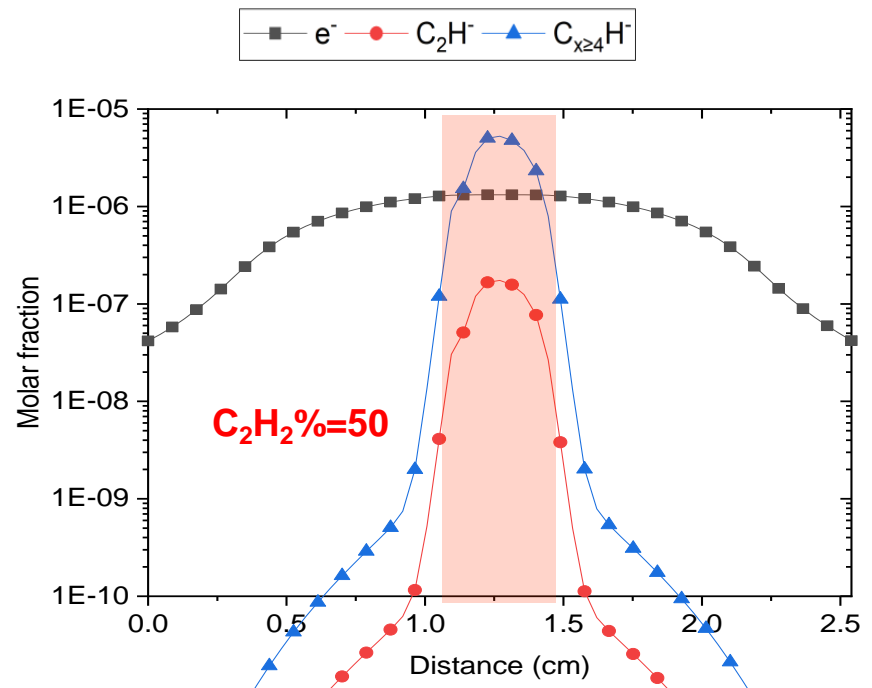
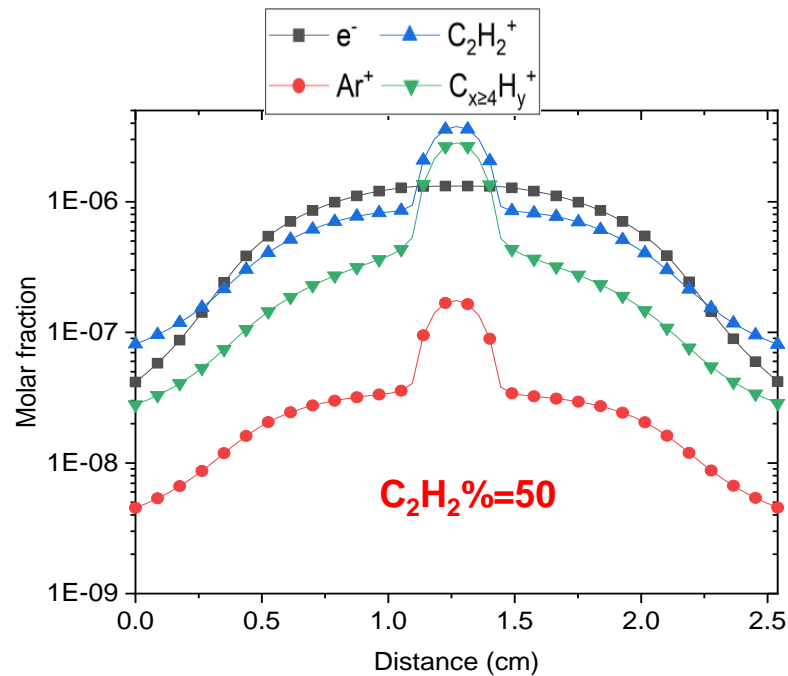


- Total acetylene conversion at low flow inlet-velocity
- Substantial conversion (50%) at large flow velocity
- Substantial production of large molecular edifice at low residence time !

G. Tetard et al. PSST (2021), **30**, paper# 105015(22 pp) doi 10.1088/1361-6595/ac2a17  
 G. Tetard et al. Submitted to Plasma Process and Polymer - 2021



# Ionization and molecular growth kinetics in Ar/C<sub>2</sub>H<sub>2</sub> CCRF plasmas : Charge species in the gap



**Narrow highly electronegative region in the center of the gap**  
**Electropositive plasma elsewhere**

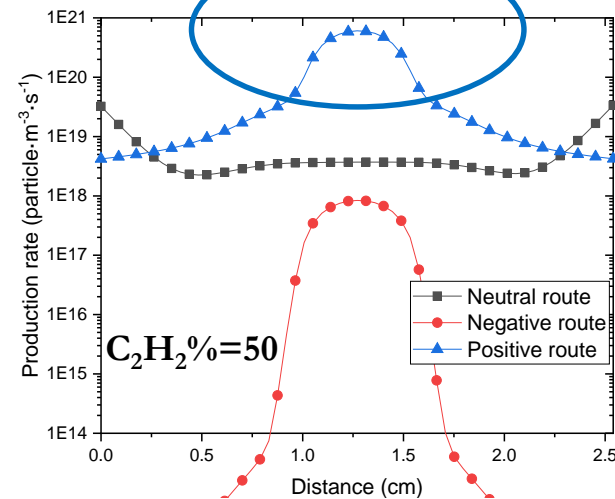
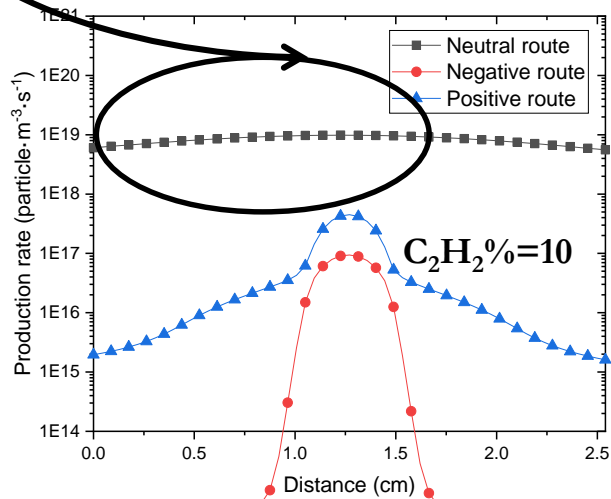
G. Tetard et al. PSST (2021), **30**, paper# 105015(22 pp) doi 10.1088/1361-6595/ac2a17

G. Tetard et al. Submitted to Plasma Process and Polymer - 2021

# Ionization and molecular growth kinetics in Ar/C<sub>2</sub>H<sub>2</sub> CCRF plasmas : major molecular growth routes

- $C_2H + C_{2n}H_2 \rightarrow C_{2n+2}H_2 + H$   $k \sim 3 \cdot 10^7 \text{ m}^3 \cdot \text{mol}^{-1} \cdot \text{s}^{-1}$
- $C_2H_2 + C_{2n}H \rightarrow C_{2n+2}H_2 + H$   $k \sim 4 \cdot 10^7 \text{ m}^3 \cdot \text{mol}^{-1} \cdot \text{s}^{-1}$
- $C_2H_2 + C_{2n}H^- \rightarrow C_{2n+2}H^- + H_2$   $k \sim 6 \cdot 10^5 \text{ m}^3 \cdot \text{mol}^{-1} \cdot \text{s}^{-1}$
- $C_2H_2 + C_{2n}H_y^+ \rightarrow C_{2n+2}H_y^+ + H_2, (n > 2, y = 6)$   $k \sim 6 \cdot 10^7 \text{ m}^3 \cdot \text{mol}^{-1} \cdot \text{s}^{-1}$

$C_{n \geq 8}H_2$  production rate



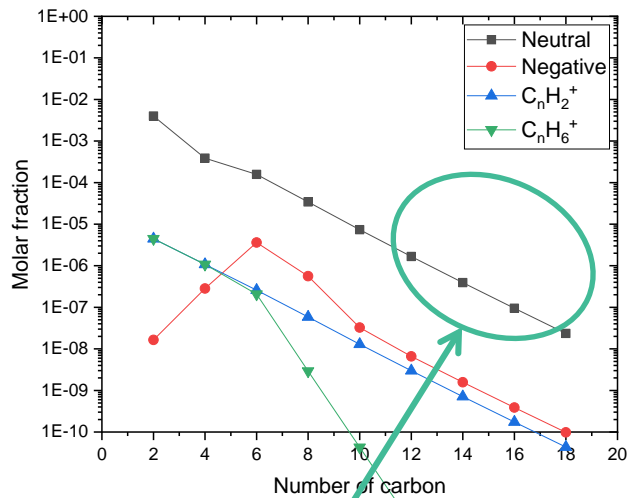
- Neutral route dominate when C<sub>2</sub>H<sub>2</sub> density is low in the gap
- Positive ion route dominate when C<sub>2</sub>H<sub>2</sub> density is significant in the gap

G. Tetard et al. Submitted to Plasma Process and Polymer - 2021

# Ionization and molecular growth kinetics in Ar/C<sub>2</sub>H<sub>2</sub> CCRF plasmas : Molecular growth and nucleation

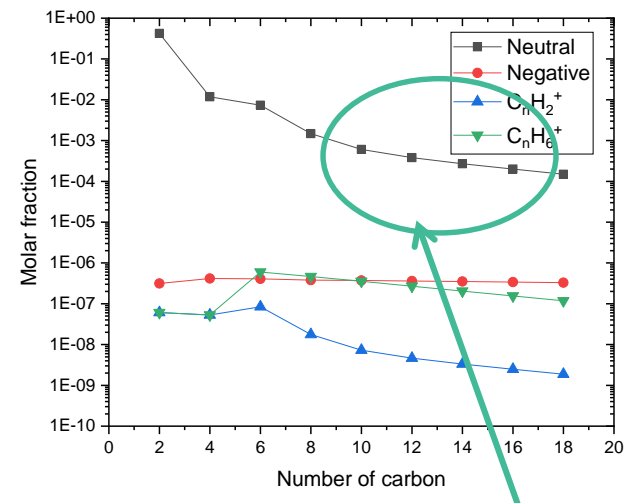
## Size distributions of hydro-carbons

### 10 % C<sub>2</sub>H<sub>2</sub> in the feed gas



Several orders of magnitude exponential population decrease with size

### 50 % C<sub>2</sub>H<sub>2</sub> in the feed gas



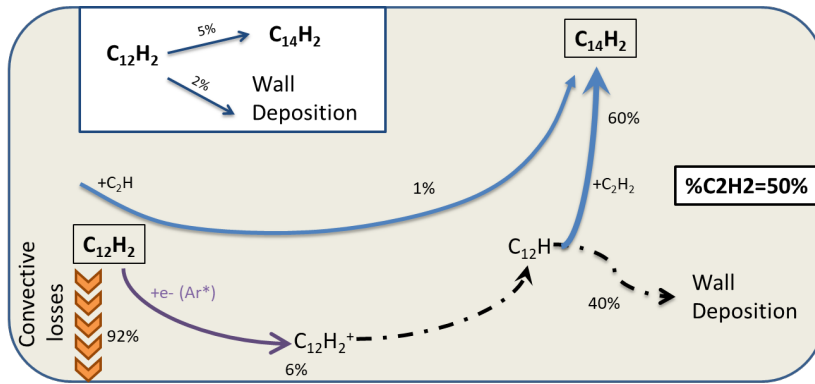
Mass accumulation over all the size distribution (very slight or no decrease)

Only the positive ion route can lead to nucleation ... in this growth model !!!

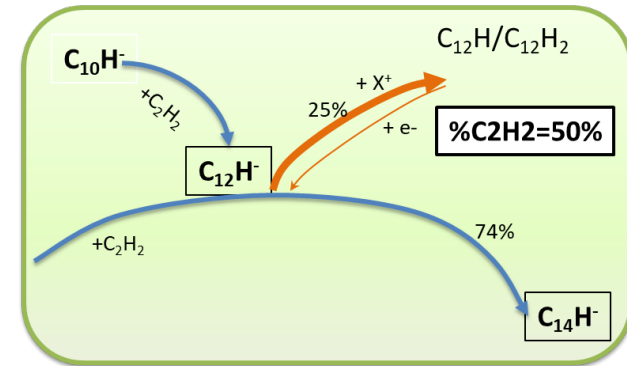
G. Tetard et al. Submitted to Plasma Process and Polymer - 2021

# Competitive processes for molecular growth

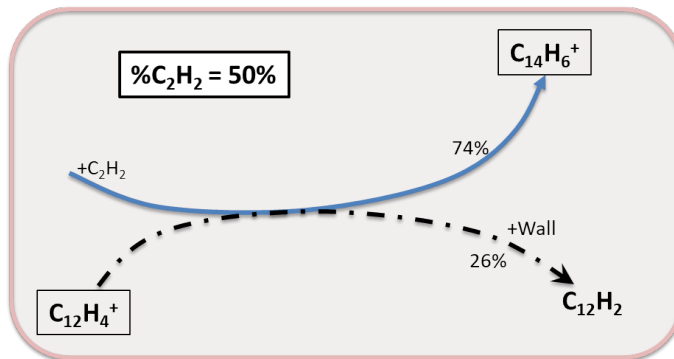
## Neutral molecule route



## Negative ion route



## Positive ion route

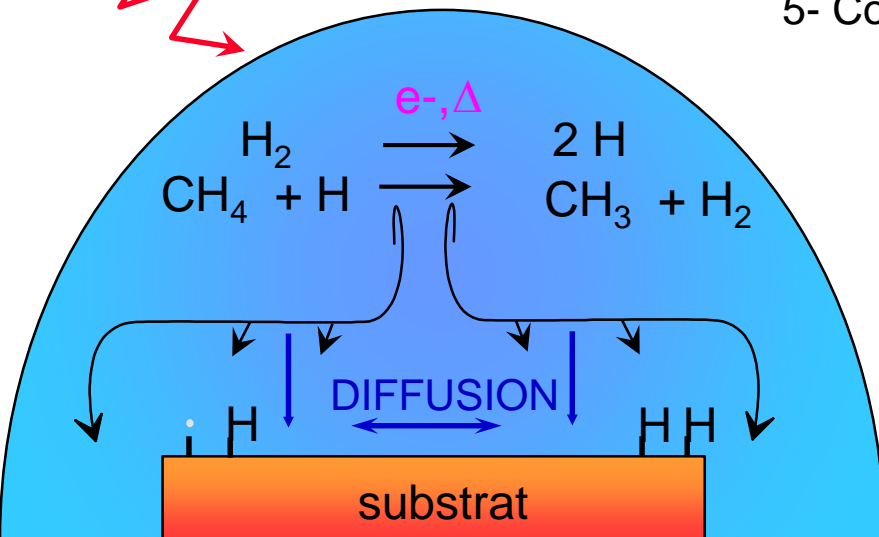
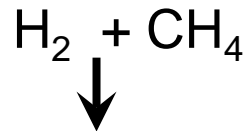


- **Convective losses of neutral clusters are highly dominant which explains the limited contribution of this route in the nucleation process**
- **Significant competition between molecular growth and wall deposition for the neutral route**
- **Molecular growth is dominant for both negative and positively charged clusters**

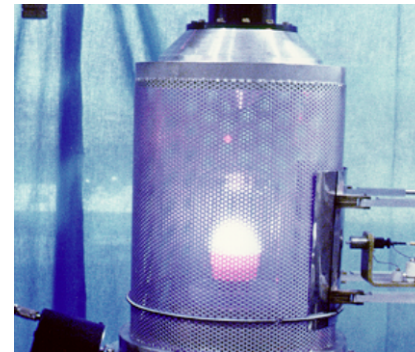
G. Tetard et al. Submitted to Plasma Process and Polymer - 2021

# Complex chemistry and the coupling between thermal, chemical and transport phenomena in hydrocarbon containing microwave plasmas

Microwave



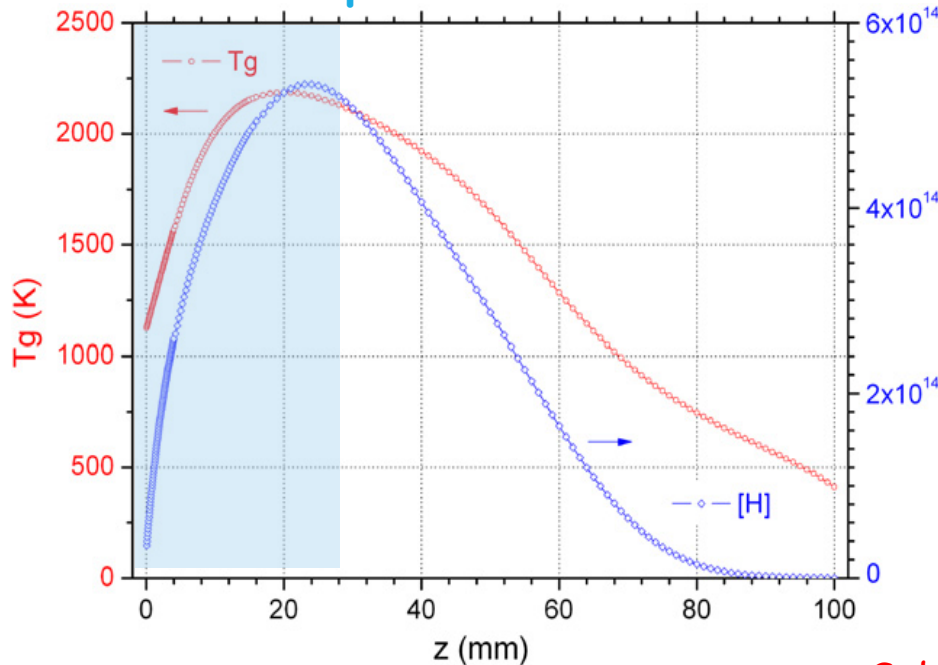
- 1- Wave-plasma interaction – energy deposition
- 2- Electron heating and collisional relaxation
- 3- Gas excitation and heating
- 4- Thermal and electron-impact chemistry
- 5- Composition change



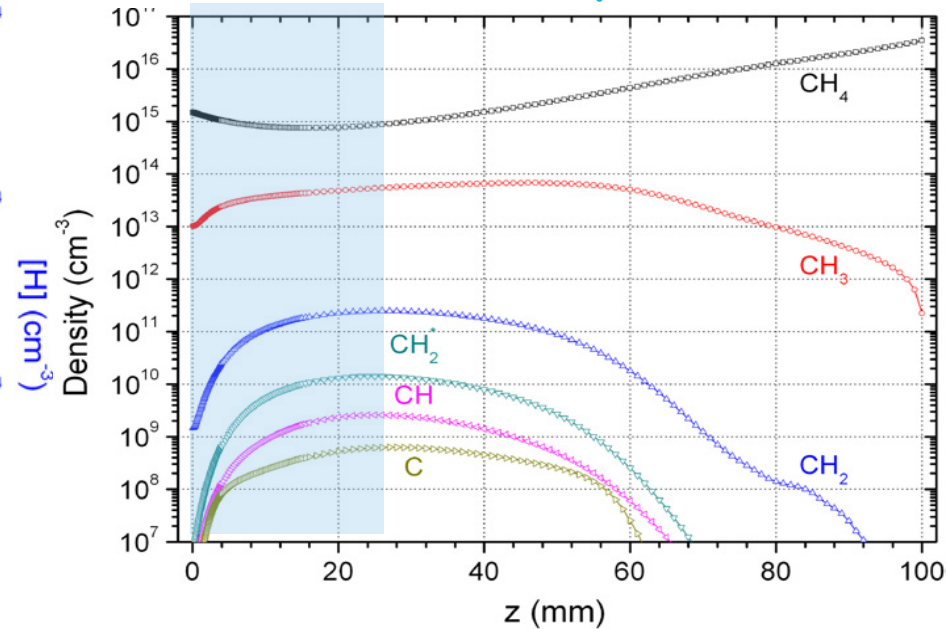
Strong coupling

# Plasma composition at low power density-low pressure (9 W/cm<sup>3</sup>, 2500 Pa)

## Gas temperature and H-atom



## 1 carbon species



Moderate temperature increase  
Electron-impact dissociation of H-atom  
Non-equilibrium H-atom density

Substantial conversion of CH<sub>4</sub>  
Smooth evolution of the plasma composition  
Slightly reactive boundary layer  
Both thermal and electron-impact chemistry

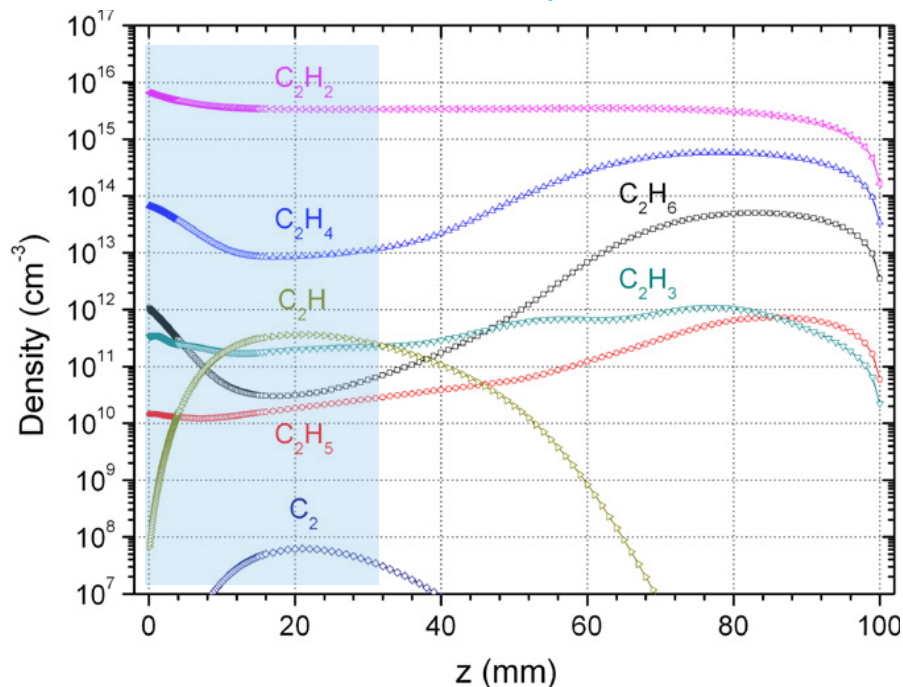
K. Hassouni et al. J. Phys. D: Appl. Phys. 43 (2010) 153001 (45pp)  
F Silva et al 2009 J. Phys.: Condens. Matter 21 364202

G. Lombardi et al. PSST 14(2005), 440-450, doi : 10.1088/0963-0252/14/3/005  
G. Lombardi et al., J. Appl. Phys. 98, 053303 (2005); <https://doi.org/10.1063/1.2034646>



# Plasma composition at low power density-low pressure (9 W/cm<sup>3</sup>, 2500 Pa)

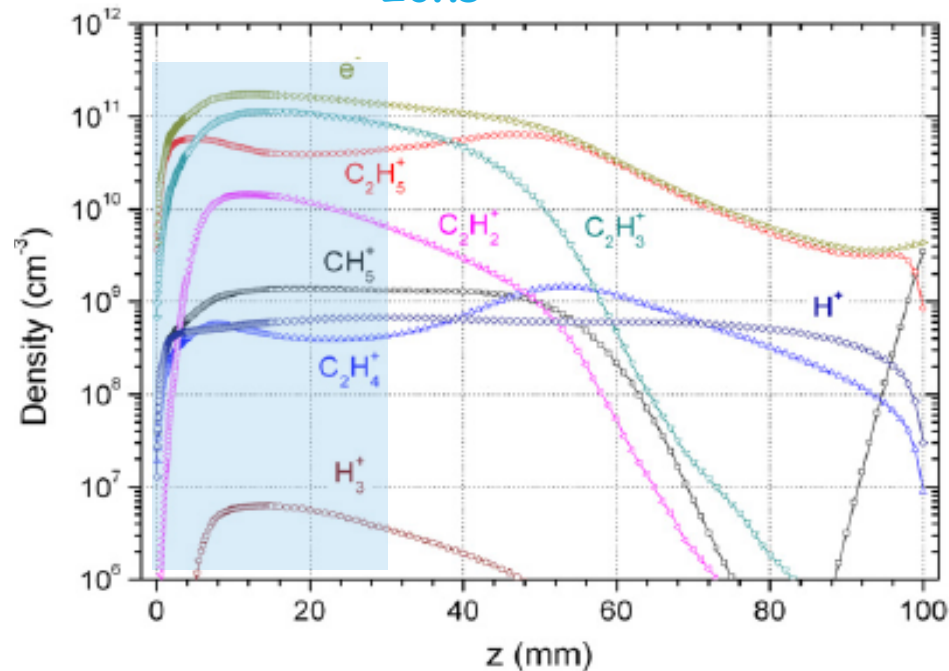
## 2 carbon species



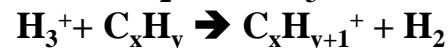
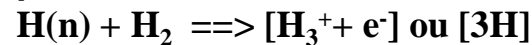
From CH<sub>4</sub> feed to Acetylene plasma  
Substantial but still 'smooth' variation of the  
Composition in the BL

K. Hassouni et al. J. Phys. D: Appl. Phys. 43 (2010) 153001 (45pp)  
F Silva et al 2009 J. Phys.: Condens. Matter 21 364202

## Ions



Dominated by hydrocarbon ions  
Hydrogen ionization followed by ion conversion  
processes

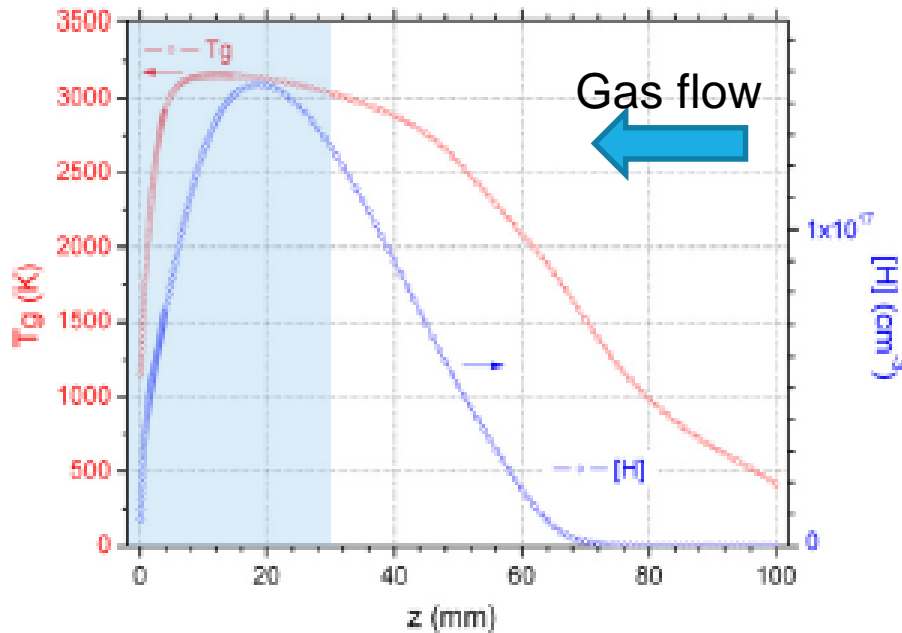


G. Lombardi et al. PSST 14(2005), 440-450, doi : 10.1088/0963-0252/14/3/005

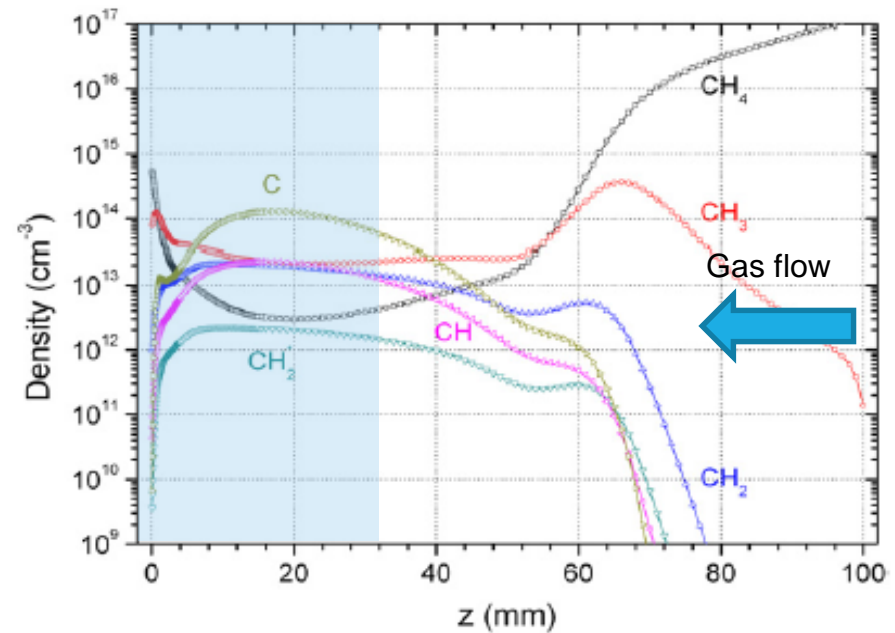
G. Lombardi et al., J. Appl. Phys. 98, 053303 (2005); <https://doi.org/10.1063/1.2034646>

# Plasma composition at high power density-high pressure (50 W/cm<sup>3</sup>, 10000 Pa)

Gas temperature and H-atom



1 carbon species



**Large temperature** increase  
**Thermal dissociation** of H-atom  
 Large dissociation yield  
**Still non-equilibrium H-atom density**

**Full conversion of CH<sub>4</sub>** before reaching the plasma bulk  
**Highly reactive boundary layer**

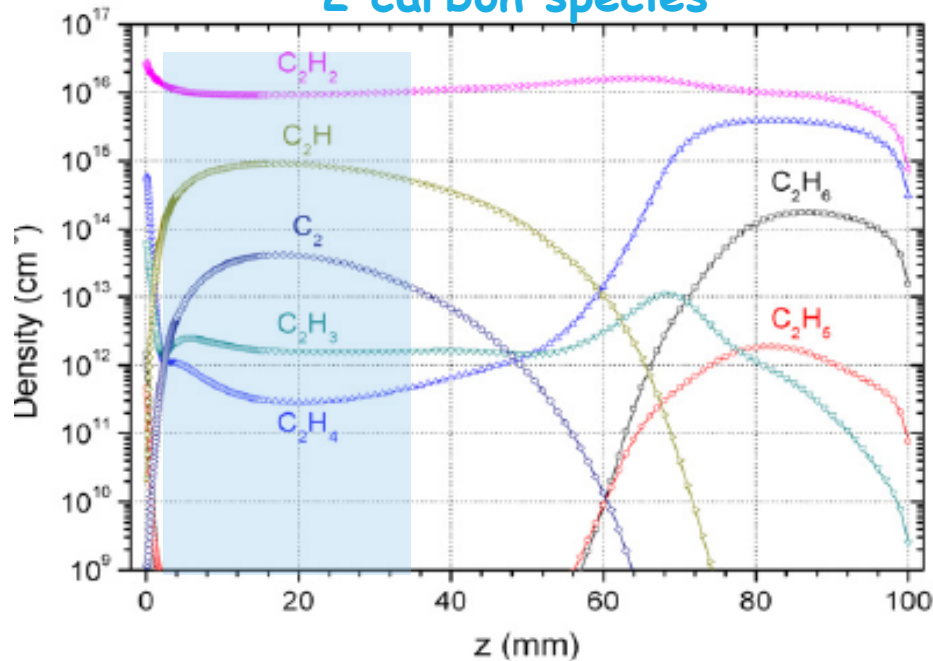
K. Hassouni et al. J. Phys. D: Appl. Phys. 43 (2010) 153001 (45pp)  
 F Silva et al 2009 J. Phys.: Condens. Matter 21 364202

G. Lombardi et al. PSST 14(2005), 440-450, doi : 10.1088/0963-0252/14/3/005  
 G. Lombardi et al., J. Appl. Phys. 98, 053303 (2005); <https://doi.org/10.1063/1.2034646>



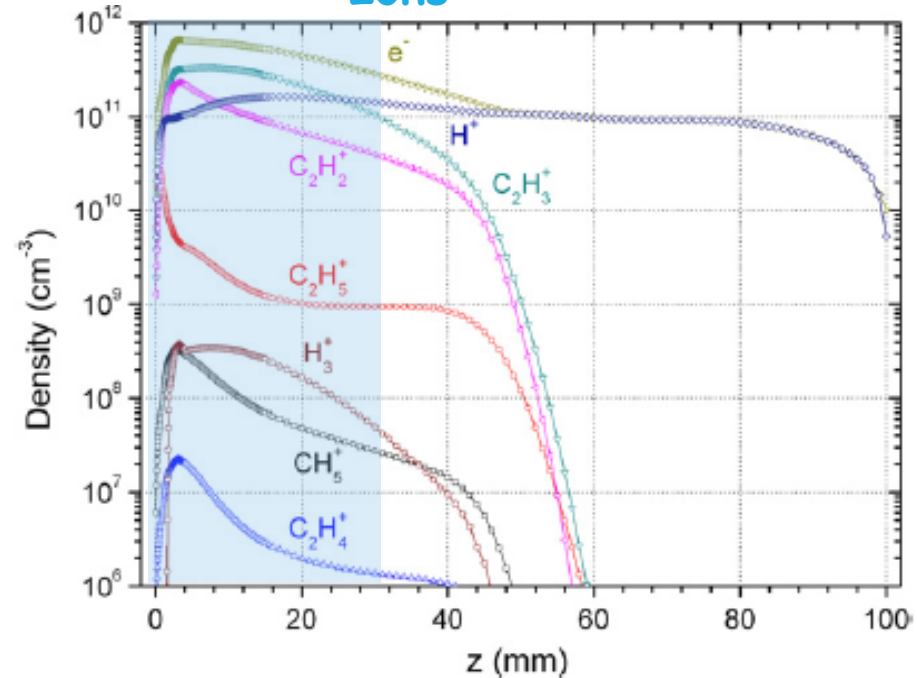
# Plasma composition at high power density-high pressure (50 W/cm<sup>3</sup>, 10000 Pa)

## 2 carbon species



Still C<sub>2</sub>H<sub>2</sub> plasma!  
Very large variation in the BL  
Driven by H-atom and T<sub>g</sub>

## Ions



H<sub>2</sub> ionization and Charge transfer in the plasma  
Substantial H-atom ionization  
Very large and highly ionized post-discharge

**Electron (microwave) → T<sub>g</sub> → H-atom THEN T<sub>g</sub> and H-atom → hydrocarbon chemistry**

K. Hassouni et al. J. Phys. D: Appl. Phys. 43 (2010) 153001 (45pp)  
F Silva et al 2009 J. Phys.: Condens. Matter 21 364202

G. Lombardi et al. PSST 14(2005), 440-450, doi : 10.1088/0963-0252/14/3/005  
G. Lombardi et al., J. Appl. Phys. 98, 053303 (2005); <https://doi.org/10.1063/1.2034646>

## Few words on the methodology (modeling)

Investigating the reactivity



Discharge physics (power coupling, electron heating)

Energy modes relaxation (energy conversion)

Species and energy transport

Interaction with the walls



We have a multi-physics problem

May addressed using several level of details/complexity

# First approach (the simplest one) : Quasi-homogeneous or plug flow plasma model

Suitable for stationary glow discharges and plasma

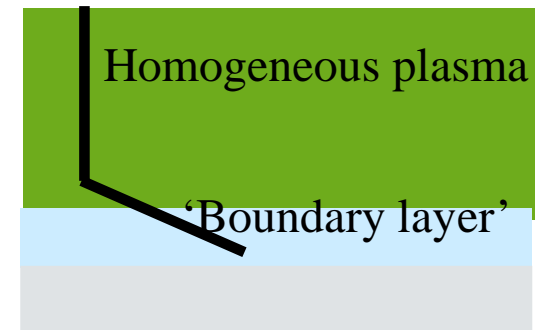
**Objective :** To get a first idea on what is happening in terms of energy relaxation and chemistry

Though it might be not accurate quantitatively.

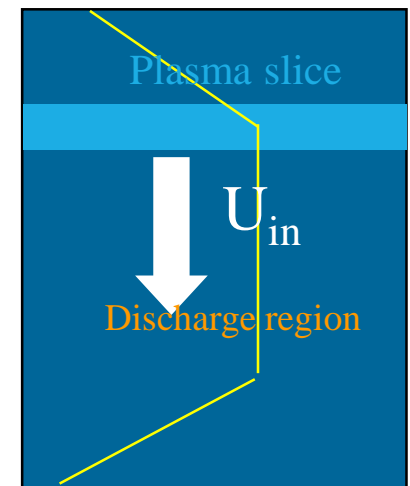
We can :

- build up very detailed CR models
- infer the major elementary processes
- setup simplified models

Quasi-homogeneous plasma

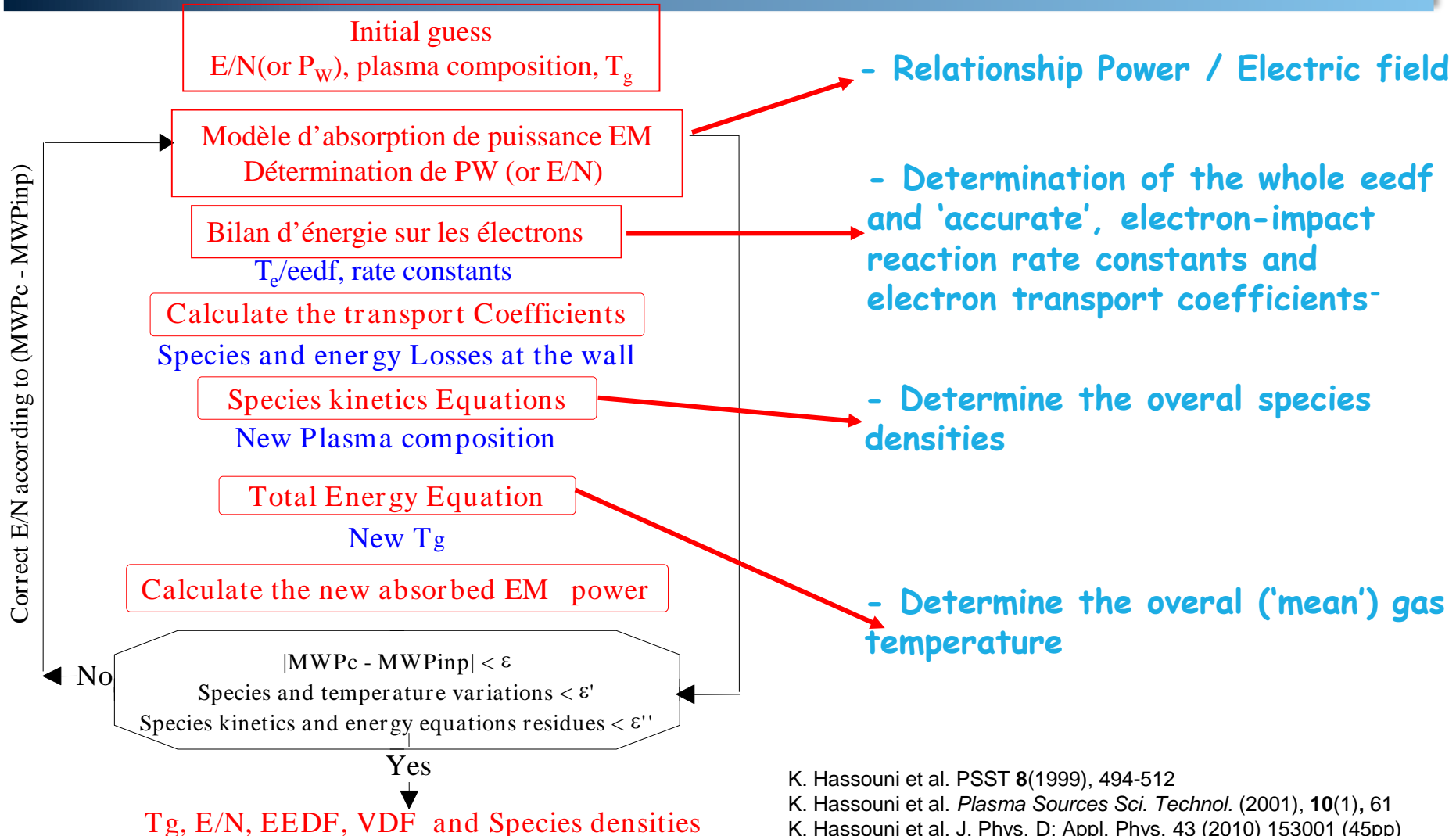


Plug flow plasma



C. D. Scott et al. JTHT(1996), **10**(3), 10.2514/3.807  
K. Hassouni et al. PSST **8**(1999), 494-512  
K. Hassouni et al. J. Phys. D: Appl. Phys. 43 (2010) 153001 (45pp)

# Typical components involved in a quasi-homogeneous plasma model



K. Hassouni et al. PSST **8**(1999), 494-512  
 K. Hassouni et al. Plasma Sources Sci. Technol. (2001), **10**(1), 61  
 K. Hassouni et al. J. Phys. D: Appl. Phys. 43 (2010) 153001 (45pp)

# Typical equations involved in quasi-homogeneous plasma model

-Relationship between power and electric field

$$P_{Ohm} = \omega_p \left( \frac{\omega_p}{v_{qm}} \right) \frac{1}{(\omega / v_{qm})^2 + 1} \frac{\epsilon_0 E_0^2}{2}$$

-Determination of the eedf

-Or the average electron energy  $f(e) = \exp(-e/kT_e)$

$$\frac{\partial f(\epsilon, t)}{\partial t} = -\frac{\partial J_E}{\partial \epsilon} - \frac{\partial J_{el}}{\partial \epsilon} - \frac{\partial J_{e-e}}{\partial \epsilon} + Q_{in} = 0$$

$$\frac{\partial \tilde{E}_e}{\partial t} = [PMW - Q_{e-v} - Q_{e-t} - Q_{e-x}] \frac{1}{\rho}$$

-Determination of the plasma composition

$$\frac{dY_s}{dt} = \frac{W_s}{\rho}$$

-Estimation of the gas temperature

$$\frac{\partial \tilde{E}}{\partial t} = [PMW - Q_{rad} - S_p] \frac{1}{\rho}$$

K. Hassouni et al. PSST **8**(1999), 494-512

K. Hassouni et al. *Plasma Sources Sci. Technol.* (2001), **10**(1), 61

K. Hassouni et al. *J. Phys. D: Appl. Phys.* 43 (2010) 153001 (45pp)

# Typical models for low pressure discharges with a local electron kinetics in the frame of the fluid approach

- X Ions in thermal equilibrium with the neutrals :  $T_i = T_g$
- X  $E = E_{\text{drift}} + E_{\text{thermal}}$  with  $E_{\text{drift}} \gg E_{\text{thermal}}$  (at least in the sheath)

A Continuity equation for each neutral, ion and electron  $\longrightarrow$  
$$\frac{\partial n_s}{\partial t} + \nabla \cdot (n_s \vec{u}_s) - W_s = 0$$

An energy equation for electrons  
Requires assumption on the eedf  $\longrightarrow$  
$$\frac{\partial n_e \langle \varepsilon_e \rangle}{\partial t} = -\nabla \cdot \left( -2/3 \kappa_e \bar{\nabla} \langle \varepsilon_e \rangle + 5/3 \bar{J}_e \langle \varepsilon_e \rangle \right) + e \bar{J}_e \cdot \bar{E} - \sum_{\text{coll}} k_{\text{coll}} n_e n \langle \varepsilon_{\text{th-coll}} \rangle$$

A momentum equation for each charged species  $\longrightarrow$  
$$\frac{\partial \rho_s \vec{u}_s}{\partial t} - \nabla \cdot (p_s) = -e \rho_s \bar{E} - \nu_{s-qm} \rho_s \vec{u}_s$$

## Simplification #1 : drift-diffusion transport

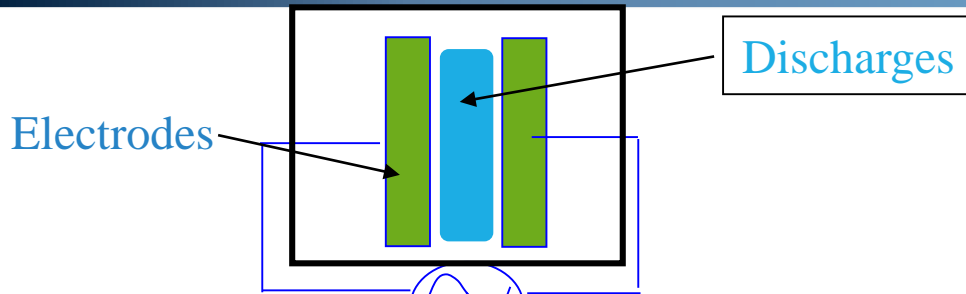
Suitable for cases with :  $\lambda_{\text{Debye}} \approx d_{\text{sys}}$  et  $\omega \ll \omega_s$

$$\frac{\partial \rho_s \vec{u}_s}{\partial t} = 0 \quad \longrightarrow \quad \vec{u}_s = z_s \mu_s \bar{E} - D_s \frac{\nabla n_s}{n_s} \quad \longrightarrow$$

(Requirfe the estimation of  $D_s$  et  $\mu_s$ )  
(advantage : no momentum equation)

**Application : Fluid model for RF ; streamer propagation, ...**

# Discharges with space charge separation : the drift-diffusion model. The example of CCRF discharges



$p=0.1-1$  torr,  $T_g=300-600$  K,  
Puissance = few 0.1 to few W

$$\lambda_{\text{Debye}} \approx 1 \text{ cm} \approx d_{\text{sys}}$$

$$V = V_0 \sin \omega t - \text{RF} - f = \omega / 2\pi = 13.56 \text{ MHz}$$

$$\omega_{p-i} \ll \omega_{\text{RF}} \ll \omega_{p-e}$$

1 Space charge is taken into account  
(some region may be non neutral)

2 Now wavelength affect :  
Electrostatic assumption

## Drift-diffusion assumption

- OK for electrons

Not valid for ions  $\Rightarrow$  ion oscillation is damped  
momentum equations required

A solution is to use : an effective electric field  
experienced by ions.

$$\frac{\partial \vec{E}_{\text{eff}}^s}{\partial t} = -v_m (\vec{E}_{\text{eff}}^s - \vec{E})$$

# Typical simulation of reactive CCRF discharges

**Module décharge**

Poisson's equation:

$$\left(\frac{\epsilon_0}{e}\right)\Delta V = n_e - \sum_{\text{ions}} z_i n_i$$

$$\frac{\partial \vec{E}_{eff}^s}{\partial t} = -\nu_m (\vec{E}_{eff}^s - \vec{E})$$

Species continuity equations:

$$\frac{\partial n_s}{\partial t} = -\nabla \cdot (\vec{\phi}_s) + S_s$$

Electron energy transport equation:

$$\frac{\partial (n_e \epsilon_e)}{\partial t} = -\nabla \cdot \vec{\phi}_e - q_e \cdot \vec{\phi}_e \cdot \vec{E} - S_e$$

**New discharge 'background gas' composition**

Solution stationnaire OR integration over long time periods

$$\frac{\partial n_s}{\partial t} = -\nabla \cdot (n_s \vec{u}_s) + W_s = 0$$

Continuity equation averaged over RF period (neutrals + ions)

**Large characteristic time species module**

**Average the rate constants and plasma properties over several RF periods**

Integration over several periods

Characteristic time smaller than RF period

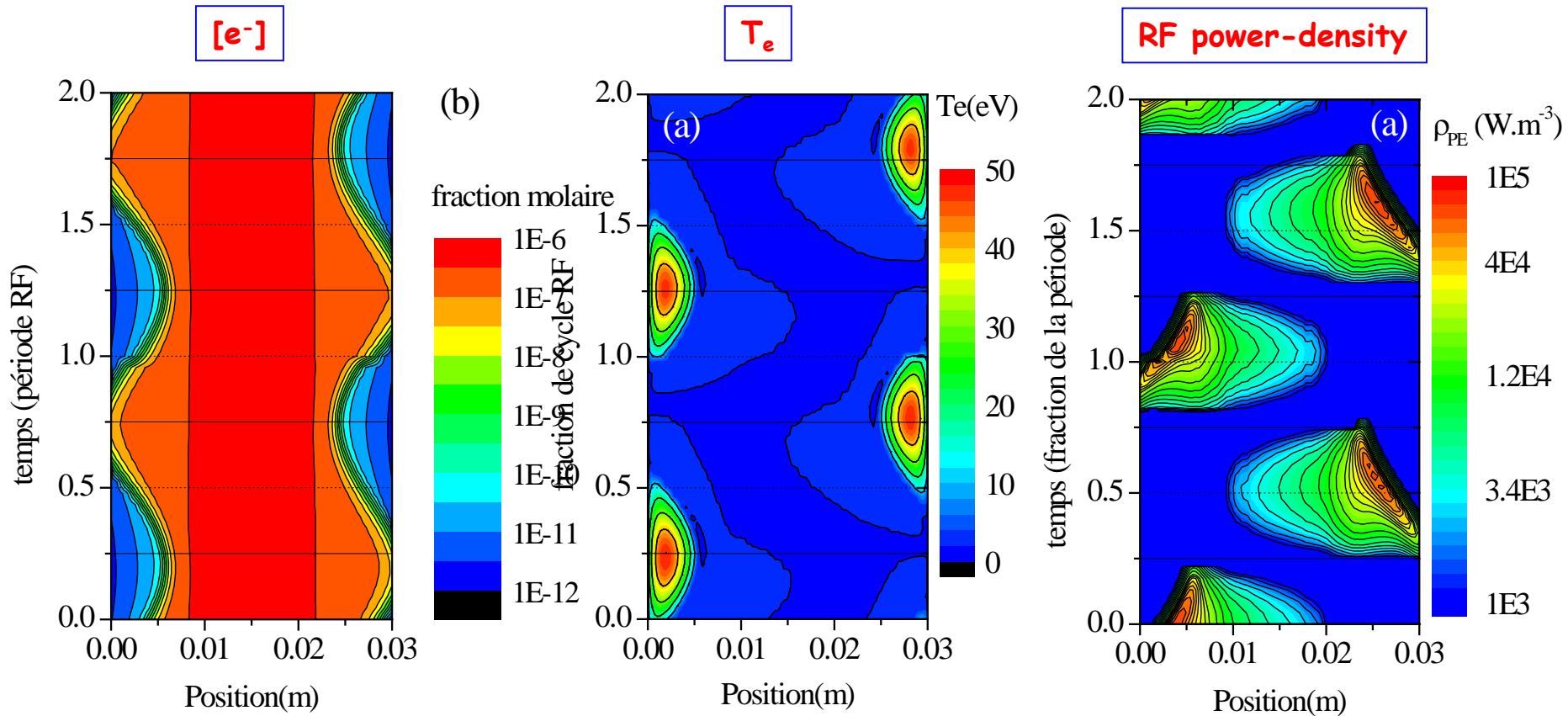
G. Tetard et al. PSST (2021), **30**, paper# 105015(22 pp) doi 10.1088/1361-6595/ac2a17

B. Kalache et al. J. Appl. Phys. (2003) **93**(6) 3198-3206

W. Morscheidt et al. Thin Solid Films (2003) **427**(1-2), pp. 219-224



# Discharge dynamics – space-time distributions of electron-mole fraction, electron-temperature and power density



$N_e$  and  $T_e$  are out of phase  
 Either very hot few electrons or a lot of warm electrons

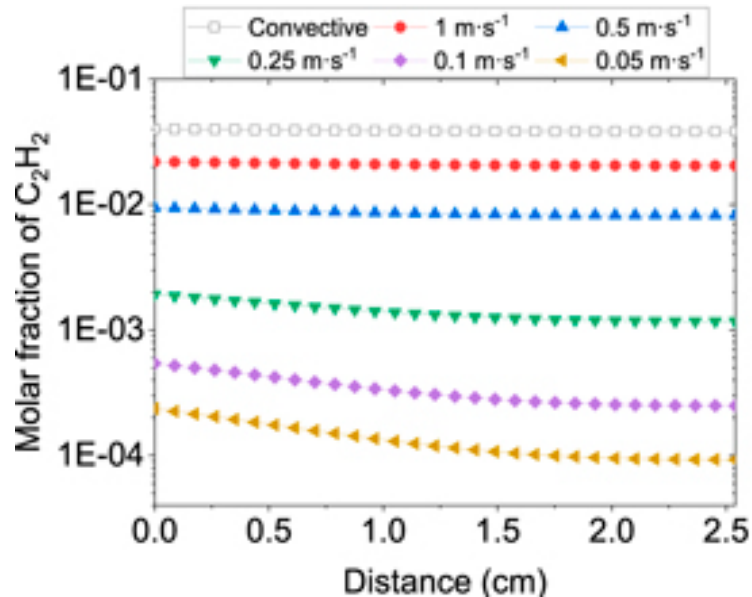
- The absorbed power is a kind of compromise between  $n_E$  and  $T_e$   
 - We need as much hot electron as possible  $\rightarrow$  the heating takes place during the anode  $\rightarrow$  cathode transition

W. Morscheidt et al. Thin Solid Films (2003) 427(1-2), pp. 219-224

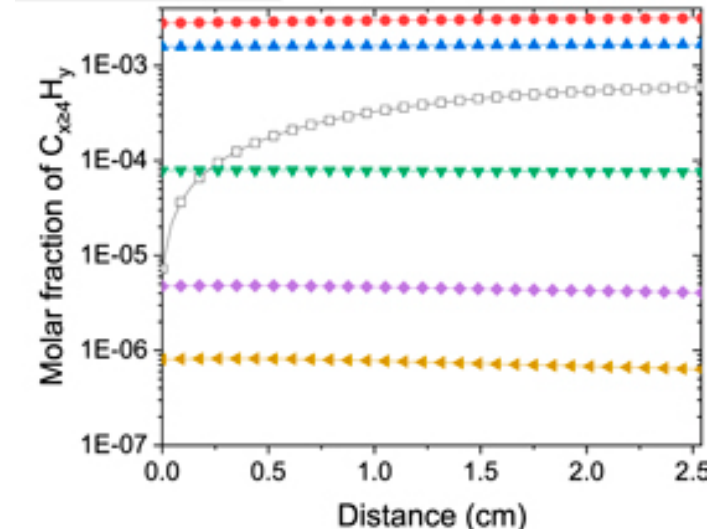
# Background gas composition

## Acetylene conversion in Ar/C<sub>2</sub>H<sub>2</sub> discharges (4% acetylene)

Acetylene fraction for different values of the inlet velocity (flow rate)



Fraction of the C<sub>2n</sub>H<sub>2</sub> (n≥4) molecules, for different values of the inlet velocity (flow rate)



The precursor may be almost totally converted and the plasma is therefore generated in a gas composition that is strongly different from the feed gas one !

G. Tetard et al. PSST (2021), **30**, paper# 105015(22 pp) doi 10.1088/1361-6595/ac2a17

G. Tetard et al. Submitted to Plasma Process and Polymer - 2021

# Transport and chemistry in low pressure plasmas

## No charge separation + fluid approach

### Simplification #2 : Ambipolar diffusion

May be used when :

$$\lambda_{\text{Debye}} \ll d_{\text{système}} \quad \longrightarrow$$

$$\sum_i z_i n_i = 0$$

$$\sum_i z_i J_i = 0$$

Special case of 1 ion + 1 électron

$$-\mu_e n_e E_{\text{amb}} - D_e \nabla n_e = \mu_i n_i E_{\text{amb}} - D_i \nabla n_i$$

$$E_{\text{amb}} = \frac{D_i - D_e}{\mu_i + \mu_e} \frac{\nabla n_e}{n_e}$$

$$J_i = -\frac{\mu_e D_i + \mu_i D_e}{\mu_i + \mu_e} \nabla n_i = -D_{\text{amb}} \nabla n_i$$

Advantage : Continuity equation = diffusion equation if no flow (no drift  $\rightarrow$  easier)

$$\frac{\partial n_i}{\partial t} + \nabla \cdot \left( -D_a \nabla n_i + \vec{u} n_i \right) - W_i = 0$$

Disadvantages : not so straightforward in the case of multicomponent plasmas (several ions , negative ions, ...)

# Multi-temperature ambipolar plasmas models

Continuity equation for each species => density  $\frac{\partial n_i}{\partial t} + \nabla \cdot (-D_i \nabla n_i + \mathbf{u} n_i) - W_i = 0$

Overall momentum => flow velocity

Electron energy =>  $T_e$   $\frac{\partial E_e}{\partial t} + \text{div}(\mathbf{u}_e h_e - \lambda_e \nabla T_e) - P_{EM} + Q_{e-t} + Q_{e-v} + Q_{e-X} = 0$

Vibration modes =>  $T_v$   $\frac{\partial E_{v-m}}{\partial t} + \text{div}(\mathbf{u}_m E_{v-m} - \lambda_{v-m} \nabla T_{v-m}) + Q_{v-t} - Q_{e-v} + Q_{v-X} = 0$

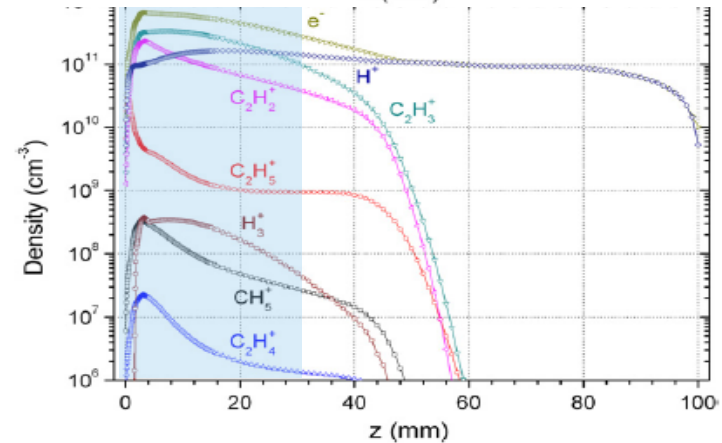
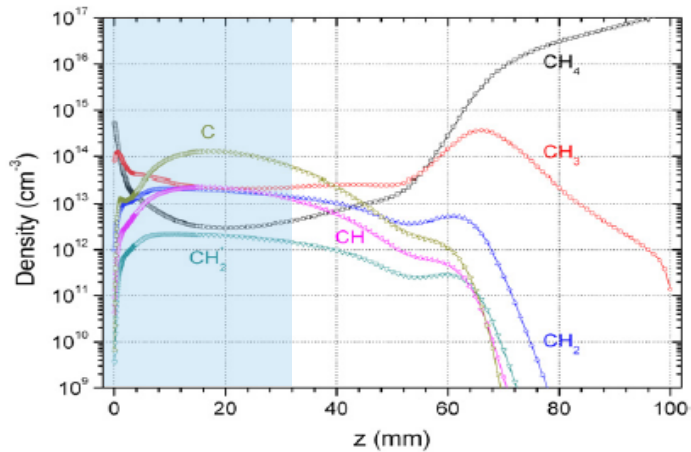
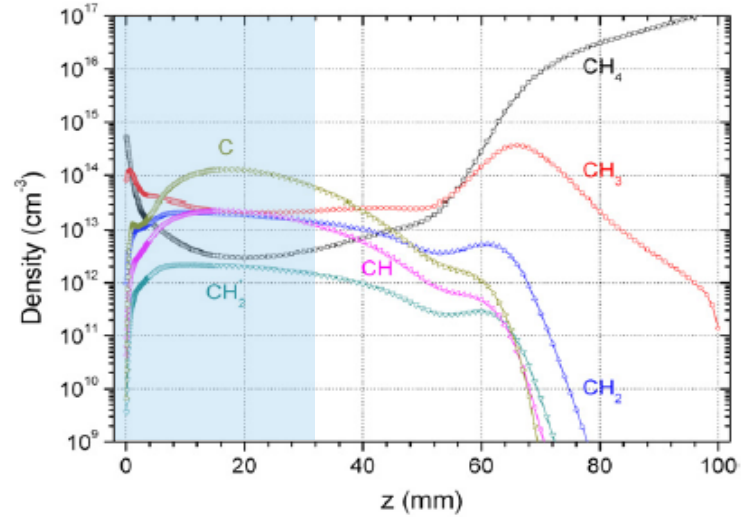
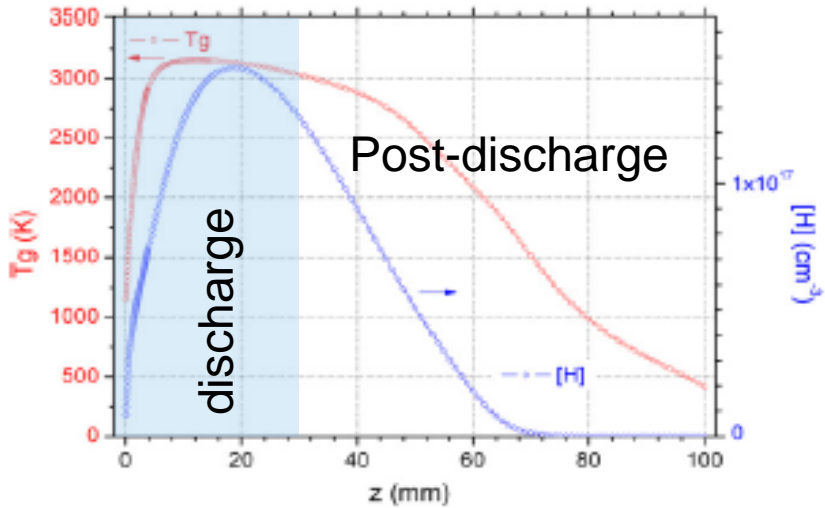
Coupling nodes

Total energy =>  $T_g$   $\frac{\partial E}{\partial t} + \text{div}[\sum \mathbf{u}_i h_i - \lambda_{t-r} \nabla T - \sum \lambda_{v-m} \nabla T_{v-m} - \lambda_e \nabla T_e] - P_{EM} + Q_{rad} = 0$

The approach is close to the one used in combustion

- K. Hassouni et al., Plasma Chem. And Plasma Proc. (1998), **18**, 325-362  
 K. Hassouni et al. J. Phys. D: Appl. Phys. 43 (2010) 153001 (45pp)  
 G. Lombardi et al. PSST **14**(2005), 440-450, doi : 10.1088/0963-0252/14/3/005  
 G. Lombardi et al., J. Appl. Phys. **98**, 053303 (2005); <https://doi.org/10.1063/1.2034646>

# Results : space distributions of temperature and species (no current, no sheath and electroneutral plasma)-example of CH<sub>4</sub>/H<sub>2</sub> microwave plasmas

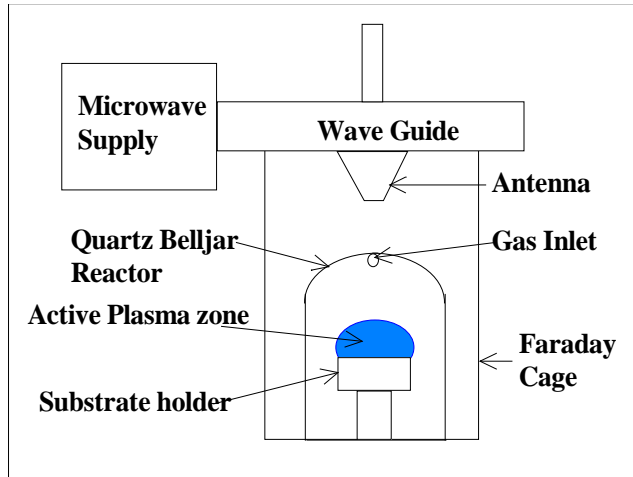


K. Hassouni et al. J. Phys. D: Appl. Phys. 43 (2010) 153001 (45pp)  
F Silva et al 2009 J. Phys.: Condens. Matter 21 364202

G. Lombardi et al. PSST 14(2005), 440-450, doi : 10.1088/0963-0252/14/3/005  
G. Lombardi et al., J. Appl. Phys. 98, 053303 (2005); <https://doi.org/10.1063/1.2034646>

# Plasma-wave interaction: The example of microwave cavity generated plasmas

Microwave coupling ,  $p=20-150$  torr,  
PMW=0.6-6 kW



Ion balance  $\implies T_e = 1-2$  eV  
power balance  $\implies 10^{11}-10^{12}$  cm $^{-3}$

$$\lambda_{\text{Debye}} \approx 0.1 \text{ mm} \ll d_{\text{sys}}$$

1 Ambipolar plasma

2  $\lambda_{\mu\text{wave}} \approx 12 \text{ cm} \approx d_{\text{sys}}$   
We have a wavelength effect:  
Quasistatic assumption not valid

3  $\omega_{p-i} \ll \omega \approx \omega_{p-e}$

Coupling with the electrons through their high frequency current that lay be excited by microwave

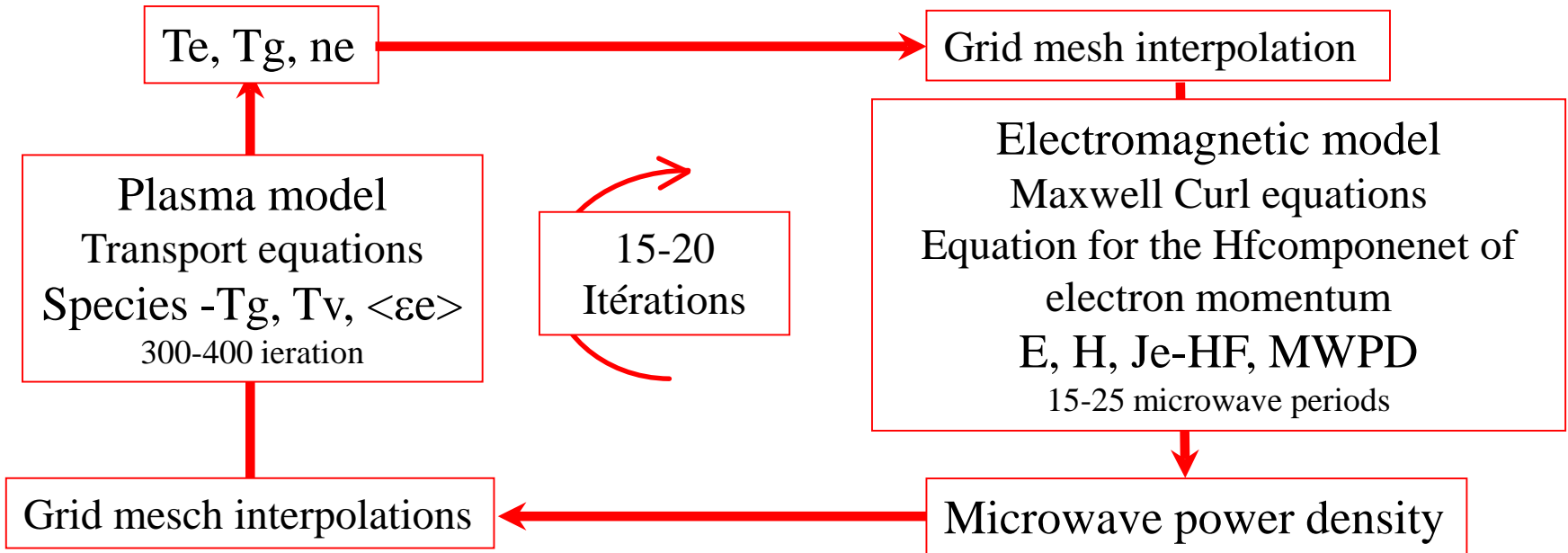
Eq. de QM-HF  
pour les  $e \implies J_{HF}$

Eq. Maxwell  $\implies E_{HF}$

$$J_{HF} \cdot E_{HF} = \text{PMW}$$

# Plasma-wave interaction: The example of microwave cavity generated plasmas

## 2. 2D self consistent model



Maxwell  
module

$$\nabla \times E = \frac{\partial H}{\partial t}$$

$$m_e \frac{d\mathbf{v}_{HF}}{dt} = -q\mathbf{E} - m_e \nu_{eff} \mathbf{v}_{HF}$$

$$\nabla \times H = J_{HF} + \epsilon_0 \frac{\partial E}{\partial t}$$

$$\mathbf{J}_{HF} = -q_e n_e \mathbf{v}_{HF}$$

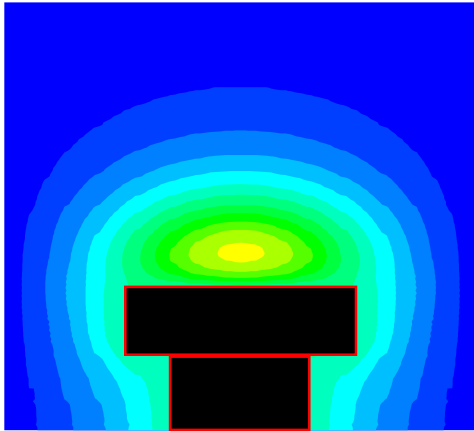
- K. Hassouni et al. J. Appl. Phys **86** (1999), 134-145.
- G. Hagelaar et al. J. Appl. Phys **96** (2004), 1819
- K. Hassouni et al. J. Phys. D. 43 (2010) 153001 (45pp)
- S. Prasanna et al. PSST 26 (2017) 097001 (6pp)
- S. Prasanna et al. PSST 25 (2016) 045017 (10pp)



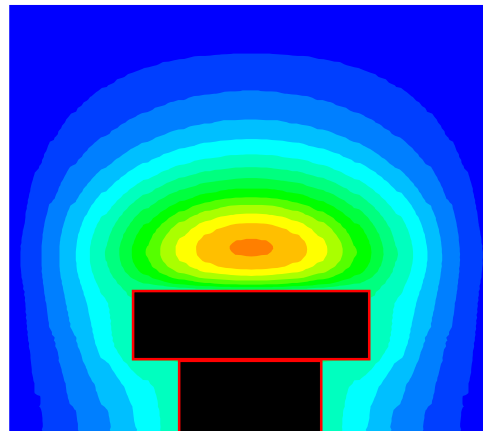
# Plasma-wave interaction. H<sub>2</sub> microwave plasma used for diamond deposition

Optimal microwave coupling – pressure = 2500 Pa

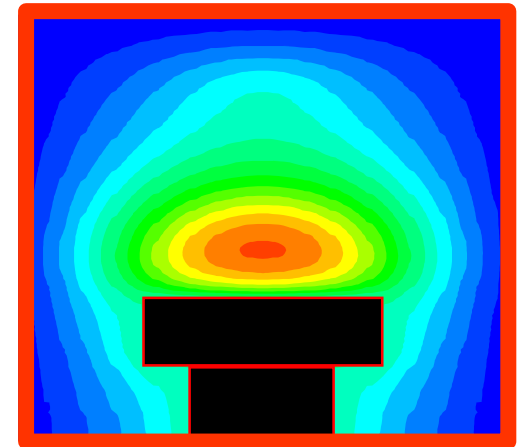
300 W



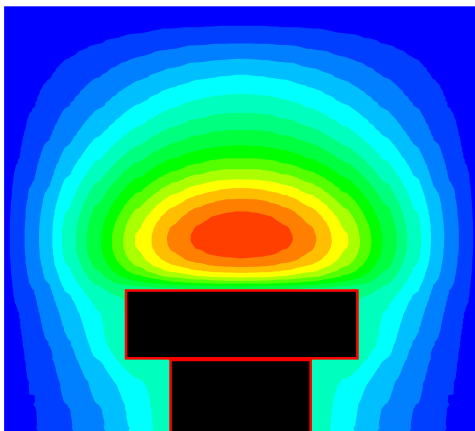
500 W



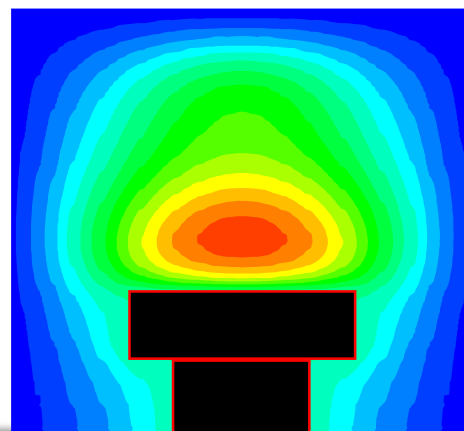
700 W



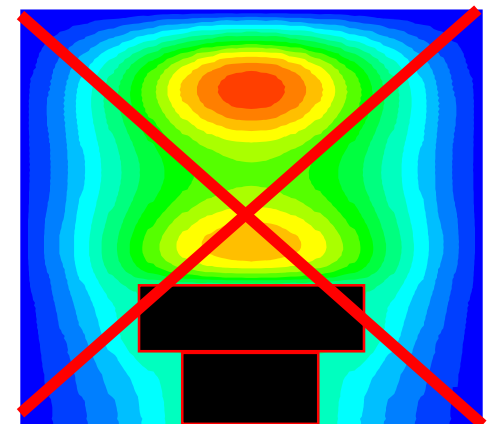
800 W



900 W

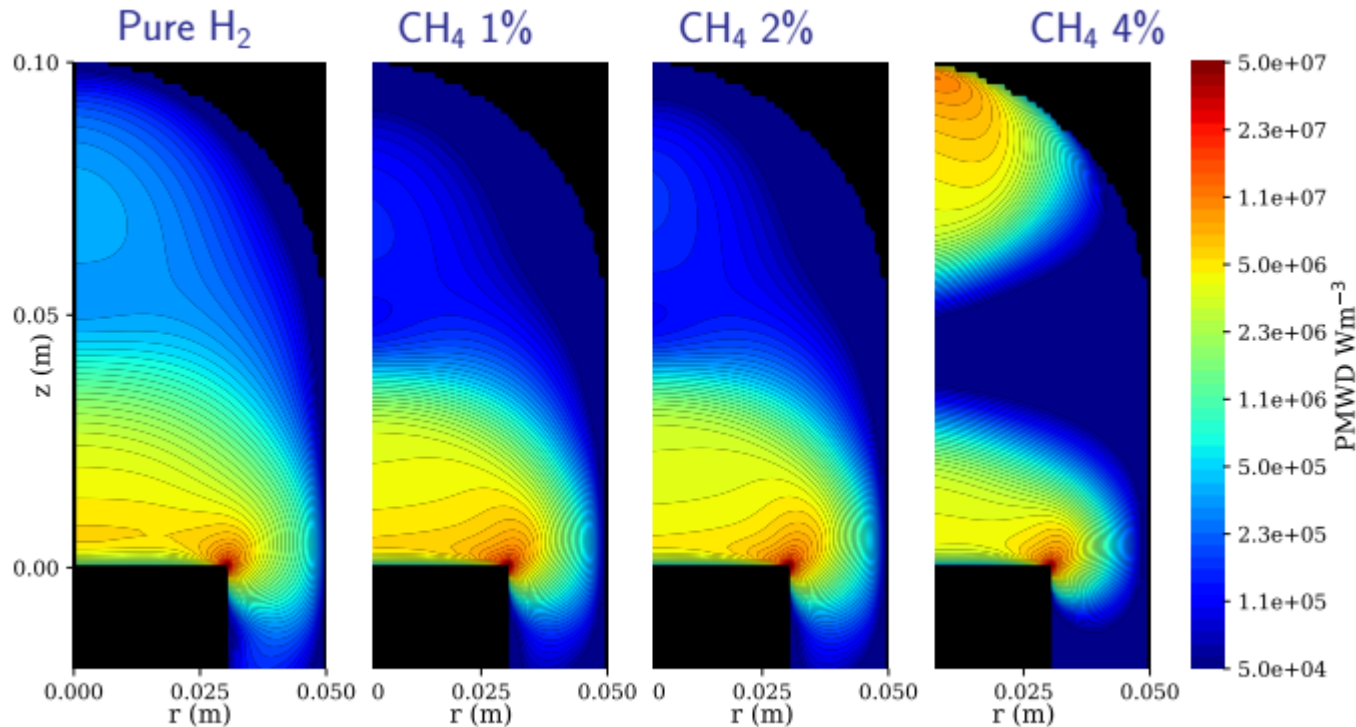


1000 W





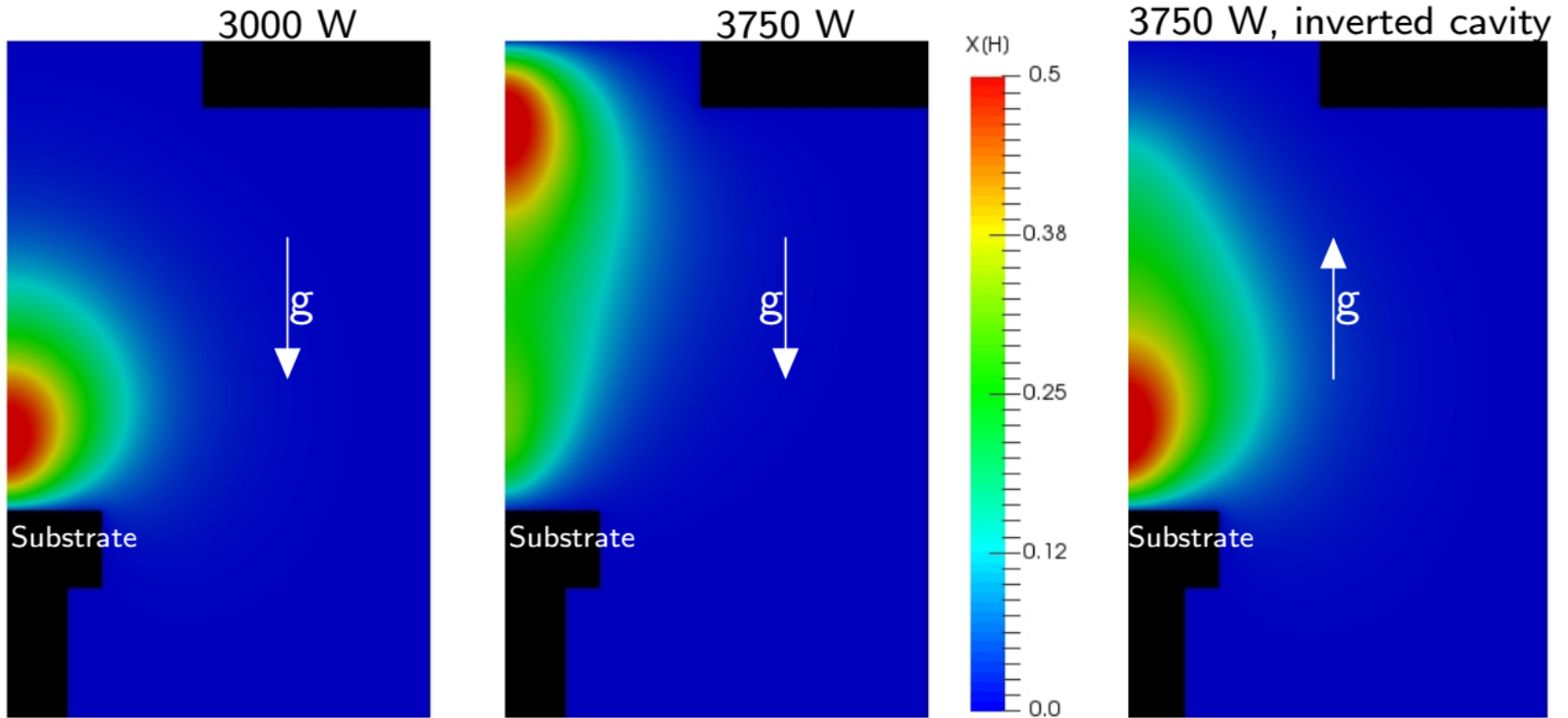
# Effect of the feed gas composition on power coupling



S. Prasanna et al. PSST 26 (2017) 097001 (6pp)

The power coupling may substantially change even with a small change in the feed gas composition

# Effect of buoyancy on power coupling



S. Prasanna et al. PSST 25 (2016) 045017 (10pp)

**The power coupling may be significantly coupled to gas flow effects  
Here we show the fact that the power coupling is very sensitive to the  
buoyancy forces and the natural convection**

# Conclusion

- Reactivity in non equilibrium plasmas : **Rich and diverse**
- Non-equilibrium effects are also **rich and diverse**
- Cannot be treated without considering discharge and plasma physics, as well as energy relaxation (molecular and atomic physics)
- Usually **strongly coupled** sets of phenomena
- Usually Huge **space and time stiffness** : key consequences on both experiments and modeling
- I did not discuss some major open questions (among others ...) :
  - Plasma Surface Interaction and its feed back on the plasma behavior
  - Collisional data issues
  - Methodological issues (still not always straightforward)
  - Multiscale approaches



# Thanks for listening

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